

## Worksheet 4.5

### Hydrolysis of Salts and Reactions of Acids and Bases

#### 4.13 Hydrolysis of Salts

Describe each as an acid, base, neutral salt, acidic salt, or basic salt. For each salt write a parent acid-base formation equation, dissociation equation, and hydrolysis equation (only for acidic and basic salts). For acids and bases write an equation to show how each reacts with water.

1.  $\text{NH}_3$       **weak base**



2.  $\text{KCl}$       **neutral salt**



3.  $\text{HNO}_3$       **strong acid**



4.  $\text{NaHCO}_3$       **basic salt**



5.  $\text{RbOH}$       **strong base**



6.  $\text{AlCl}_3$       **acid salt**



# Solutions Acids And Bases Wordwise Answers

**Annelies Wilder-Smith**

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