

Acids and bases review sheet Answer key

1. List the differences between an acid and a base. Your description should include such things as whether both are electrolytes, which contains a hydrogen ion and which includes a hydroxide ion, and you should also include two examples of an acid and two examples of a base that are present in your household.

Acid	base
Conducts electricity Contains hydrogen ion Proton donor Turns litmus paper red React with most metals Releases hydronium Ex: lemon Vinegar Battery acid (sulfuric acid)	conducts electricity contains hydroxide ion proton acceptor turns litmus paper blue releases hydroxide ion Ex: Windex bleach soap

2. Listed below are some of the properties of acids and bases. Fill in the blanks with the appropriate word, **acids** or **bases**:

- a) **A** produce hydrogen ions (H^+) in solution
- b) **B** have a bitter taste
- c) **B** produce hydroxide ions (OH^-) ions
- d) **A** react with bases to form salts
- e) **A** have a sour taste
- f) **B** have a slippery, soapy feel
- g) **B** react with acids to form salts
- h) **A** react with many metals

3. Write an equation for the dissociation of the following **acids** in water: follow example

- a) $\text{HClO}_4 \rightarrow \text{H}^+ + \text{ClO}_4^-$
- b) $\text{H}_2\text{SO}_4 \rightarrow 2\text{H}^+ + \text{SO}_4^{2-}$
- c) $\text{HC}_2\text{H}_3\text{O}_2 \rightarrow \text{H}^+ + \text{C}_2\text{H}_3\text{O}_2^-$
- d) $\text{H}_2\text{S} \rightarrow 2\text{H}^+ + \text{S}^{2-}$
- e) $\text{HCl} \rightarrow \text{H}^+ + \text{Cl}^-$
- f) $\text{HNO}_3 \rightarrow \text{H}^+ + \text{NO}_3^-$

4. Write an equation for the dissociation (ionization) of the following **bases** in water: (follow example)

- a) $\text{Mg}(\text{OH})_2 \rightarrow \text{Mg}^{2+} + 2\text{OH}^-$
- b) $\text{NaOH} \rightarrow \text{Na}^+ + \text{OH}^-$
- c) $\text{Ba}(\text{OH})_2 \rightarrow \text{Ba}^{2+} + 2(\text{OH}^-)$
- d) $\text{KOH} \rightarrow \text{K}^+ + \text{OH}^-$
- e) $\text{Mg}(\text{OH})_2 \rightarrow \text{Mg}^{2+} + 2(\text{OH}^-)$
- f) $\text{LiOH} \rightarrow \text{Li}^+ + \text{OH}^-$

5) A. What type of reaction is a **neutralization** reaction? **Double replacement**

B. Write the net equation for neutralization:



Solutions And Acids Bases Test Review 2014 Answers

Lisa M. Stone, AACN



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Table of Contents Solutions And Acids Bases Test Review 2014 Answers

1. Understanding the eBook Solutions And Acids Bases Test Review 2014 Answers
 - The Rise of Digital Reading Solutions And Acids Bases Test Review 2014 Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Solutions And Acids Bases Test Review 2014 Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Solutions And Acids Bases Test Review 2014 Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Solutions And Acids Bases Test Review 2014 Answers
 - Personalized Recommendations
 - Solutions And Acids Bases Test Review 2014 Answers User Reviews and Ratings
 - Solutions And Acids Bases Test Review 2014 Answers and Bestseller Lists

5. Accessing Solutions And Acids Bases Test Review 2014 Answers Free and Paid eBooks
 - Solutions And Acids Bases Test Review 2014 Answers Public Domain eBooks
 - Solutions And Acids Bases Test Review 2014 Answers eBook Subscription Services
 - Solutions And Acids Bases Test Review 2014 Answers Budget-Friendly Options
6. Navigating Solutions And Acids Bases Test Review 2014 Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Solutions And Acids Bases Test Review 2014 Answers Compatibility with Devices
 - Solutions And Acids Bases Test Review 2014 Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Solutions And Acids Bases Test Review 2014 Answers
 - Highlighting and Note-Taking Solutions And Acids Bases Test Review 2014 Answers
 - Interactive Elements Solutions And Acids Bases Test Review 2014 Answers
8. Staying Engaged with Solutions And Acids Bases Test Review 2014 Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Solutions And Acids Bases Test Review 2014 Answers
9. Balancing eBooks and Physical Books Solutions And Acids Bases Test Review 2014 Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Solutions And Acids Bases Test Review 2014 Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Solutions And Acids Bases Test Review 2014 Answers
 - Setting Reading Goals Solutions And Acids Bases Test Review 2014 Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Solutions And Acids Bases Test Review 2014 Answers
 - Fact-Checking eBook Content of Solutions And Acids Bases Test Review 2014 Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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