## CHAPTER 1 CHEMISTRY: THE STUDY OF CHANGE

#### Problem Categories

#### Difficulty Level

Zigori 1.5, 1.11, 3.13, 4.14, 3.13, 1.21, 1.22, 1.23, 3.24, 3.25, 3.36, 1.29, 3.36, 4.34, 4.32, 4.33, 1.34, 4.36,

AMOREGON: S.A. T. T. S. T. T. S. T. DOL. T. DO

\$6000 min 1.50, 1.60, 1.60, 1.60, 1.60, 1.60, 1.60, 1.60, 1.60, 1.60, 1.60, 1.60, 1.60, 1.100

1.3	60	Quantities vo. This statement clearly investors a measurable distance.
	86.5	Qualitative. This is a value palgroom. There is no numerical scale of incommon for artistic excellence.
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	1989	Qualitative: Another rather indement.
	100	Qualitative. Example transform and becarbook they are not the second of measurement.
1:4	0.6	Repaintments (Re) have (RE) theoretic
1.11	***	Character's property. Degree gas is commend to a construction reaction, to composition and identity are chargest.
	(0.5	Chemical property. The fractions is commoned by the greeting plants; it is besself into vegetable metter intellered compactions.
	469	Physical property. The encamenment of the building point of mater does not change to identity or acceptablish.
	649	Physical property. The measurement of the densities of lead and absorbers does not change their companion.
	444	Observabled property. When minimum undergoes machine decay, the products are characteristic different substances.
1.13	190	Physical change. The holium tor's changed in any may by traking out of the bullous.
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	360	Physical change. The orange paice concentrate can be represented by evaporation of the water.
	100	-Characterist change. Photosymbosis changes water, codem distribut, etc., later complex organic matter

46). Physical change. The naticula by reservoired anothergod by evoporation.

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