



ACIDS & BASES 1

- 1) a) What is a base? (1)
b) What is an alkali? (1)
c) What is the reaction between an acid and a base called? (1)
d) Give two everyday examples of reactions between acids and bases.
1
2 (2)

- 2) a) Match the descriptions below to the pH's in this list: 2 5 7 9 13 (2)

description	neutral	very alkaline	slightly acidic	slightly alkaline	very acidic
pH					

- b) i) Litmus indicator could not distinguish between solutions of pH 2 and 5. Explain why. (1)
ii) Name an indicator that could. (1)

- 3) Complete the following word equations: .
a) zinc + nitric acid \rightarrow (2)
b) hydrochloric acid + copper carbonate \rightarrow (2)
c) sulfuric acid + sodium oxide \rightarrow (2)
d) potassium hydroxide + sulfuric acid \rightarrow (2)
e) hydrochloric acid + lithium carbonate \rightarrow (2)
f) magnesium oxide + nitric acid \rightarrow (2)
g) sulfuric acid + strontium \rightarrow (2)
h) potassium carbonate + sulfuric acid \rightarrow (2)
i) rubidium hydroxide + hydrochloric acid \rightarrow (2)

- 4) a) Why should you not add an alkali to an acid spill? (2)
b) Give two safer ways of treating acids spills.
1
2 (2)

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Decoding **Task 1chemsheets Acids And Bases Worked Solutions**: Revealing the Captivating Potential of Verbal Expression

In a period characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its power to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Task 1chemsheets Acids And Bases Worked Solutions**," a mesmerizing literary creation penned by way of a celebrated wordsmith, readers embark on an enlightening odyssey, unraveling the intricate significance of language and its enduring affect our lives. In this appraisal, we shall explore the book is central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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