

When $k_1 = k_2$ the (Counterintuitive?) result is that the same size of MFR and PFR will maximize the concentration of R (the intermediate product): $C_{R, \max}$ is higher in the PFR, but the volumes are the same!

My solution (partly graphical) showed that $V_{PFR} \approx 153 \text{ L}$ while $V_{MFR} \approx 167 \text{ L}$. In fact, they should be identical! ^(?)

Here is the justification (after Kelly Caputo pointed it out to me... Thanks, Kelly!!!):

$$\tau_{MFR, \text{opt}} = \frac{1}{\sqrt{R_1 R_2}} = \underline{10 \text{ min (OK)}} \quad \frac{V_{PFR} = V_{MFR} = 167 \text{ L}}{\nearrow}$$

(eq. 8.15)

For the PFR, we use L'Hôpital's theorem

$$\tau_{PFR, \text{opt}} = \lim_{R_2/k_1 \rightarrow 1} \frac{\ln\left(\frac{R_2}{R_1}\right)}{R_2 - R_1} = \frac{\lim\left(1 + \ln \frac{R_2}{R_1}\right)}{\lim(R_2)} = \frac{1}{R_2} = \underline{10 \text{ min}}$$

So, the important comment here, in agreement with Levenspiel's statement on p.176, is NOT ONLY that the MFR is "not much worse than" the PFR, But that in terms of activity - even though conversion is lower - the MFR does not have the typical size disadvantage as well. (vs. selectivity)

Please add this page to your [HW #6] Solution package!

Prentice Hall Inc Chemistry Practice Problems Answers

Michael Hoelscher



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web may 13 2009 welcome everyone it is a pleasure to be here with you today to help celebrate international nurses day

and to applaud the great work done by new zealand nurses in so many different and challenging settings both here and overseas