

Name: KEY

Pr. _____

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Write balanced formula equations for these reactions and identify the type of reaction.

1. Potassium + Iodine	$2K + I_2 \rightarrow 2KI$	synthesis
2. Barium Iodide + Sodium Sulfate	$BaI_2 + Na_2SO_4 \rightarrow BaSO_4 + 2NaI$	double rep.
3. Aluminum Hydroxide	$2Al(OH)_3 \rightarrow Al_2O_3 + 3H_2O$	decomposition
4. Sodium Chlorate	$NaClO_3 \rightarrow NaCl + O_2$	decomposition
5. Aluminum Bromide + Chlorine	$AlBr_3 + Cl_2 \rightarrow AlCl_3 + Br_2$	single replacement
6. Phosphoric Acid + Magnesium	$H_3PO_4 + Mg \rightarrow Mg_3(PO_4)_2 + H_2$	single rep.
7. Iron III Sulfate + Calcium Nitrate	$Fe_2(SO_4)_3 + 3Ca(NO_3)_2 \rightarrow 2Fe(NO_3)_3 + 3CaSO_4$	double rep.
8. Sodium + Water	$2Na + 2H_2O \rightarrow 2NaOH + H_2$	single rep.
9. Aluminum + Iodine	$2Al + 3I_2 \rightarrow 2AlI_3$	synthesis
10. Iron III Hydroxide + Hydrobromic Acid	$Fe(OH)_3 + 3HBr \rightarrow FeBr_3 + 3H_2O$	double rep.
11. Magnesium Hydroxide + Ammonium Sulfate	$Mg(OH)_2 + (NH_4)_2SO_4 \rightarrow MgSO_4 + 2NH_4OH$	double rep.
12. Water	$\frac{electricity}{2H_2O} \rightarrow 2H_2 + O_2$	decomposition
13. Calcium + Hydrochloric Acid	$Ca + 2HCl \rightarrow CaCl_2 + H_2$	single replacement
14. Lithium + Chlorine	$2Li + Cl_2 \rightarrow 2LiCl$	synthesis

Ws 4 Balancing Equations And Predicting Products

Karin Nielsen-Saines

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