

1 • Chemistry Foundations

Name _____

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Units and Measurements

a. SI Units (see Appendix 1 of textbook)

Mass (kg) – the property that resists acceleration

| SI units | Common | Conversion |
|------------------------|-------------|----------------|
| Length | meter | m |
| Mass | kilogram | kg |
| Time | second | s |
| Volume (1 liter) | cubic meter | m ³ |
| Temperature | Kelvin | K |
| Amount of Substance | mole | mol |

Example 1: Mass – units for other physical quantities that are proportional to Mass (force)

Example: What is the mass of a 1 kg mass that is accelerated at 10 m/s²?

$$\text{mass} = \frac{\text{force}}{\text{acceleration}} = \boxed{0.1 \text{ kg}}$$

Example: Energy is measured in Joules (J). Kinetic energy is the energy of moving objects. For the

equation $E_k = \frac{1}{2}mv^2$, a mass of 1 kg is accelerated at 10 m/s².

$$\text{Energy} = \frac{1}{2}mv^2 = \frac{1}{2}(1 \text{ kg})(10 \text{ m/s})^2$$

$$E_k = \frac{1}{2}(1 \text{ kg}) \left[\frac{\text{m}^2}{\text{s}^2} \right] = \boxed{5 \text{ J}}$$

Example: Specific heat is $\frac{\text{Joules}}{\text{kg} \cdot ^\circ\text{C}}$. Water has a specific heat of 4.18 J/kg °C.

$$\frac{\text{mass} \cdot \text{specific heat} \cdot \Delta T}{\Delta T} = \boxed{1.67 \text{ kg}}$$

b. Metric Prefixes – Adapt your SI units system. The metric system is composed of several units that are related to each other by powers of 10. Memorize the following prefixes and their abbreviations.

| Prefix | Symbol | Factor |
|--------|--------|-------------------|
| kilo | k | 10 ³ |
| hecto | h | 10 ² |
| deka | da | 10 ¹ |
| deci | d | 10 ⁻¹ |
| centi | c | 10 ⁻² |
| milli | m | 10 ⁻³ |
| micro | μ | 10 ⁻⁶ |
| nan | n | 10 ⁻⁹ |
| pico | p | 10 ⁻¹² |
| femto | f | 10 ⁻¹⁵ |
| atto | a | 10 ⁻¹⁸ |
| zepto | z | 10 ⁻²¹ |
| yocto | y | 10 ⁻²⁴ |

Example: 1000 g is 1 kg.

$$1000 \text{ g} = \frac{10^3 \text{ g}}{10^3 \text{ g/kg}} = \boxed{1 \text{ kg}}$$

Example: 1000 s is 1 h.

$$1000 \text{ s} = \frac{10^3 \text{ s}}{3600 \text{ s/h}} = \boxed{0.278 \text{ h}}$$

Example: 1000 kg is 1 t.

$$1000 \text{ kg} = \frac{10^3 \text{ kg}}{1000 \text{ kg/t}} = \boxed{1 \text{ t}}$$

Example: A unit for volume is a liter (L). The abbreviation is **liters**.

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America's leading physical chemist and another of Hale's inspired acquisitions took a former trade school and forged from it a grandiose university among the orange groves. It would be impossible while reading Goodstein's lively account not to be impressed by the energy drive and boundless enthusiasm of men like Millikan, Hale, and Noyes who had the bare-faced audacity to set about building an institute to rival the cream of the universities of Europe and America. Marcus Chown, *New Scientist*: Goodstein's story is first and foremost the tale of three men: the astronomer George Ellery Hale, the chemist Alfred Noyes, and the physicist Robert Millikan. It is the story of their attempts to transform an undistinguished little school founded in 1891 into a world-class scientific establishment. A useful book. Tony Rothman, *Science*: In *Millikan's School*, the story of Throop University's transformation into Caltech is told with precision. Judith Goodstein's history offers a quick tour of the landmarks of science in the mid-20th century and a glance at how pure science puts itself at the service of government, commerce, and the military. Goodstein approaches her subject with a healthy sense of humor and an acute sense of academic politics. She tells a wonderful story about how Caltech lost to Princeton in a bidding war over the services of Albert Einstein, for example. To her credit, Goodstein asks the hard question: What is the best way to do science? *Millikan's School* offers enough hard data to enable us to come to our own conclusions. Jonathan Kirsch, *Los Angeles Times*: A cleanly written, scientifically well-informed account of one of the world's foremost institutions for science and technology. Ed Regis, *Nature*: Relying on archival material, published secondary sources, and interviews with institute scientists, Goodstein presents a highly readable account of Caltech's beginnings at the turn of the century, substantive, informative, and a good read. Rebecca S. Lowen, *Technology and Culture*: As a history of science, this book is well-crafted. Orderly in its flow, it is not only a tribute to Millikan but also places him within the development of physics as a field. Andrew Rolle, *Southern California Quarterly*: A fascinating history that speaks to issues far larger than Cal Tech itself. This well-written and honest account witnesses the many cited instances of anti-Semitism in the scientific world; it is both a good read and a sobering reminder that big science and top schools are not brought by storks. Carroll Pursell, *History of Education Quarterly*: The author focuses on the personalities and the research fields of the principal scientific figures. The emphasis on personalities and capsule surveys of relevant scientific fields produce a book that can be apprehended by a wide audience. Roger Geiger, *Isis*: This chronicle offers glimpses of the passion and drive that have motivated a roster of distinguished scientists. *Publishers Weekly*: A lively tale. Goodstein's individual profiles are lean and candid; her background on subjects as diverse as nuclear astrophysics, seismology, aeronautical design, quantum mechanics, and rocket fuel are crisp and understandable. With a light style and meticulous documentation, Goodstein has produced a tale worthy of her subject. Marshall Robinson, *Foundation News*: A distinguished and uniquely American institution has found its chronicler, and its chronicle, in Judith Goodstein's thorough but compact story of *Millikan's School*. The emergence of Caltech as a powerhouse of science and engineering and a makeweight in the technological advancement of 20th-century industry is both beautifully and reliably presented. Harry Woolf, *Institute for*

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