

## Chemical Formula Writing Worksheet Solutions

Write chemical formulas for the compounds in each box. The names are found by finding the intersection between the cations and anions. Example: The first box is the intersection between the "zinc" cation and the "chloride" anion, so you should write "ZnCl<sub>2</sub>", as shown.

	<i>zinc</i>	<i>iron (II)</i>	<i>iron (III)</i>	<i>gallium</i>	<i>silver</i>	<i>lead (IV)</i>
<i>chloride</i>	<b>ZnCl<sub>2</sub></b>	<b>FeCl<sub>2</sub></b>	<b>FeCl<sub>3</sub></b>	<b>GaCl<sub>3</sub></b>	<b>AgCl</b>	<b>PbCl<sub>4</sub></b>
<i>acetate</i>	<b>Zn(C<sub>2</sub>H<sub>3</sub>O<sub>2</sub>)<sub>2</sub></b>	<b>Fe(C<sub>2</sub>H<sub>3</sub>O<sub>2</sub>)<sub>2</sub></b>	<b>Fe(C<sub>2</sub>H<sub>3</sub>O<sub>2</sub>)<sub>3</sub></b>	<b>Ga(C<sub>2</sub>H<sub>3</sub>O<sub>2</sub>)<sub>3</sub></b>	<b>Ag C<sub>2</sub>H<sub>3</sub>O<sub>2</sub></b>	<b>Pb(C<sub>2</sub>H<sub>3</sub>O<sub>2</sub>)<sub>4</sub></b>
<i>nitrate</i>	<b>Zn(NO<sub>3</sub>)<sub>2</sub></b>	<b>Fe(NO<sub>3</sub>)<sub>2</sub></b>	<b>Fe(NO<sub>3</sub>)<sub>3</sub></b>	<b>Ga(NO<sub>3</sub>)<sub>3</sub></b>	<b>AgNO<sub>3</sub></b>	<b>Pb(NO<sub>3</sub>)<sub>4</sub></b>
<i>oxide</i>	<b>ZnO</b>	<b>FeO</b>	<b>Fe<sub>2</sub>O<sub>3</sub></b>	<b>Ga<sub>2</sub>O<sub>3</sub></b>	<b>Ag<sub>2</sub>O</b>	<b>PbO<sub>2</sub></b>
<i>nitride</i>	<b>Zn<sub>3</sub>N<sub>2</sub></b>	<b>Fe<sub>3</sub>N<sub>2</sub></b>	<b>FeN</b>	<b>GaN</b>	<b>Ag<sub>3</sub>N</b>	<b>Pb<sub>3</sub>N<sub>4</sub></b>
<i>sulfate</i>	<b>ZnSO<sub>4</sub></b>	<b>FeSO<sub>4</sub></b>	<b>Fe<sub>2</sub>(SO<sub>4</sub>)<sub>3</sub></b>	<b>Ga<sub>2</sub>(SO<sub>4</sub>)<sub>3</sub></b>	<b>Ag<sub>2</sub>SO<sub>4</sub></b>	<b>Pb(SO<sub>4</sub>)<sub>2</sub></b>

Write the formulas for the following compounds:

- 1) copper (II) chloride **CuCl<sub>2</sub>**
- 2) lithium acetate **LiC<sub>2</sub>H<sub>3</sub>O<sub>2</sub>**
- 3) vanadium (III) selenide **V<sub>2</sub>Se<sub>3</sub>**
- 4) manganese (IV) nitride **Mn<sub>3</sub>N<sub>4</sub>**
- 5) beryllium oxide **BeO**
- 6) sodium sulfate **Na<sub>2</sub>SO<sub>4</sub>**
- 7) aluminum arsenide **AlAs**
- 8) potassium permanganate **KMnO<sub>4</sub>**
- 9) chromium (VI) cyanide **Cr(CN)<sub>6</sub>**
- 10) tin (II) sulfite **SnSO<sub>3</sub>**
- 11) vanadium (V) fluoride **VF<sub>5</sub>**
- 12) ammonium nitrate **NH<sub>4</sub>NO<sub>3</sub>**

# Writing Chemical Formula Answers

**Disha Experts**

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the 'Disha Experts' text.

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### **Table of Contents Writing Chemical Formula Answers**

1. Understanding the eBook Writing Chemical Formula Answers
  - The Rise of Digital Reading Writing Chemical Formula Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Writing Chemical Formula Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Writing Chemical Formula Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Writing Chemical Formula Answers
  - Personalized Recommendations



- Writing Chemical Formula Answers User Reviews and Ratings
- Writing Chemical Formula Answers and Bestseller Lists
- 5. Accessing Writing Chemical Formula Answers Free and Paid eBooks
  - Writing Chemical Formula Answers Public Domain eBooks
  - Writing Chemical Formula Answers eBook Subscription Services
  - Writing Chemical Formula Answers Budget-Friendly Options
- 6. Navigating Writing Chemical Formula Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Writing Chemical Formula Answers Compatibility with Devices
  - Writing Chemical Formula Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Writing Chemical Formula Answers
  - Highlighting and Note-Taking Writing Chemical Formula Answers
  - Interactive Elements Writing Chemical Formula Answers
- 8. Staying Engaged with Writing Chemical Formula Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Writing Chemical Formula Answers
- 9. Balancing eBooks and Physical Books Writing Chemical Formula Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Writing Chemical Formula Answers
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Writing Chemical Formula Answers
  - Setting Reading Goals Writing Chemical Formula Answers
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Writing Chemical Formula Answers
  - Fact-Checking eBook Content of Writing Chemical Formula Answers

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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