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**Patrick R. Delaney, O. Ray Whittington**



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introduced to basic accounting principles and practice and then led into more complex issues of pricing costing financial analysis budgeting and financial projections The third tier will explore future developments in sport finance and funding and examine how they will impact on sport management knowledge and skill requirements Within this framework the book will provide students with two types of learning outcomes The first outcome is knowledge based and will include the following a grounded understanding of the commercial evolution of sport from 1950 the present an understanding of the different commercial phases sport goes through to reach commercial maturity an appreciation of the major global sport events and leagues and their financial arrangements an insight into the financial strengths and weaknesses of contemporary sport The second outcome is skill based and by reading the book and completing the cases students will be able to identify the different legal structures of sport organisations and the financial implications of each structure explain the main accounting conventions and how they impact on the financial management of sport organisations identify the foundations of double entry bookkeeping construct a simple set of accounts for a sport organisation use financial ratio analysis to evaluate the financial performance of sport organisations understand different methods of constructing budgets and use them to control the financial operation of a sport organisation set up models for identifying and managing costs identify different methods for setting prices for sport goods and services and to apply them to specific settings and events undertake a financial planning exercise and or feasibility study for a sport organisation or facility undertake a cost benefit analysis for a sport organisation event Therefore students will obtain both a broad knowledge of the commercial evolution of sport and its current financial operation AND the necessary skills for understanding and supervising the financial operations of a sport organisation

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exams and care is taken to ensure that they cover all the information candidates need to pass the CPA Exam *Audit and Accounting Guide: Investment Companies* AICPA, 2018-11-06 Whether you are a financial statement preparer or auditor it is critical to understand the complexities of the specialized accounting and regulatory requirements for investment companies This 2018 guide provides authoritative how to accounting and auditing advice including implementation guidance and illustrative financial statements and disclosures This guide is the industry standard resource supporting practitioners in a constantly changing industry landscape packed with continuous regulatory developments Updates include References to appropriate AICPA Technical Questions and Answers that address when to apply the liquidation basis of accounting Appendices discussing the new standards for financial instruments leases and revenue recognition Appendices discussing common or collective trusts and business development companies CPA Examination Review Irvin N. Gleim, 1997

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for the nitration 4 01 ml of nitric acid 4 00 ml of sulfuric acid and 4 56 ml bromobenzene were used to complete the reaction with these measurements bromobenzene was found to be the limiting reagent since only 4 56 ml of bromobenzene was used the amount of product predicted to be recovered was 8 7533 g

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the aim of the experiment is to determine and study electrophilic aromatic substitution reaction eas particularly nitration of methyl benzoate go give methyl m nitrobenzoate thus determine the weight melting point and percentage yield of the pure

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in this experiment we will carry out the nitration of bromobenzene bromine substituted on an aromatic ring is an ortho para directing group it will direct an incoming electrophile to the ortho and para positions it is also deactivating meaning that its reactions will be slower than the identical reactions of benzene

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section 20711 lab experiment nitration of bromobenzene introduction the chemical reaction of nitration consists of a nitro group being added to or substituted in a molecule nitration can basically be carried out by a mixture of concentrated nitric acid and sulfuric acid this mixture is useful to obtain the active nitronium ion

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1 go to the restroom and wash the affected area with lots of soap and water the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area 3 the summary to this warning is not to touch any part of your body while performing this experiment if you must leave the lab for any reason 1

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in the nitration of bromobenzene a mixture of sulfuric acid and nitric acid react with one another nitric acid the weaker of the two acids will serve as base to form nitronium ion bromobenzene will attack the nitronium ion by nucleophilic attack

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then actual is your actual findings yield actual theoretical  $\times 100$  what are the possible side reactions of nitration of bromobenzene dinitration can occur at ortho para or at both ortho positions occurs in overheating meta product happens very rarely because bromine is an ortho para director

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the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area theory and explanation of the nitration of bromobenzene if the temperature of the reaction mixture exceeds 60 o

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the bromine substituent in bromobenzene is an ortho para director the para position is favored due to steric effects in the reaction that takes place in this lab a nitro group is substituted onto bromobenzene at the para position to form 1 bromo 4 nitrobenzene

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study with quizlet and memorize flashcards containing terms like reacting sulfuric and nitric acid together produces the nitronium ion alone nitric acid is not a strong enough electrophile to react with bromobenzene sulfuric acid is added to the nitric acid the reverse rxn can overheat and boil easily includes a claisen adaptor and

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over 5 minutes slowly pipet this bromobenzene into the icy cold nitration mixture hno 3 h 2 so 4 in the 125 ml flask and stir continue to stir the reaction mixture in the ice bath for 30 minutes do not heat in a new 250 ml erlenmeyer

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on the right the arenium cation has a nitro set with incredibly small 0 6 angstrom bond distances this will result in an optimization failure as the atoms are even close for the programme to accomodate their electrons demonstration of creating the p arenium cation by to nitration of bromobenzene

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the purpose of this lab was to observe where substituents are added on a benzene ring by way of electrophilic aromatic substitution the electrophilic source in the nitration of bromobenzene comes from a nitronium ion no2 which acts as the electrophilic species

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nitration of bromobenzene is an electrophilic aromatic substitution reaction the electrophile will be formed from 4 0ml of nitric acid and 4 0ml of sulfuric acid and will then react with 4 5ml of bromobenzene this will produce o bromonitrobenzene and p bromonitrobenzene

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