

### Post-Lab Questions

1. Based on your X-Y graph, which one of  $\text{Na}_2\text{CO}_3$  or  $\text{H}_2\text{SO}_4$  is the limiting reactant in each of the seven flasks in the experiments?

Flask#	1	2	3	4	5	6	7
Limiting Reactant							

2. Based on your experimental results, what is the molar ratio of  $\text{CO}_2$  (a product) and  $\text{Na}_2\text{CO}_3$  (a reactant)?

3. If you react 1.205 g of  $\text{Na}_2\text{CO}_3$  with 1.205 g of  $\text{H}_2\text{SO}_4$ , how many grams of  $\text{CO}_2$  will be produced.

# Stoichiometry By Loss Of Co<sub>2</sub> Lab Answers

**J Spring**



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