

Stoichiometry Worksheet and Key



1. How many moles of O_2 will be formed from 1.65 moles of KClO_3 ?

1.65 mol KClO_3	_____ mol O_2	= _____ mol O_2
	_____ mol KClO_3	

2. How many moles of KClO_3 are needed to make 3.50 moles of KCl ?

3.50 mol KCl		= _____ mol KClO_3

3. How many moles of KCl will be formed from 2.73 moles of KClO_3 ?

_____		= _____



4. How many moles of Fe_2O_3 are produced when 0.275 moles of Fe is reacted?

0.275 mol Fe		= _____ mol Fe_2O_3

5. How many moles of Fe_2O_3 are produced when 31.0 moles of O_2 is reacted?

_____		= _____

6. How many moles of O_2 are needed to react with 8.9 moles of Fe ?

_____		= _____



7. How many moles of O_2 are produced when 1.26 moles of H_2O is reacted?

8. How many moles of H_2O are needed to produce 55.7 moles of H_2 ?

9. If enough H_2O is reacted to produce 3.40 moles of H_2 , then how many moles of O_2 must have been made?
(a bit challenging, but just think about it and you can probably figure it out)

Stoichiometry Sheet 2 Answers

O. Jerry Parker, Gary L. Breneman

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the authors' names.

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Improving Student Comprehension of Stoichiometric Concepts Connie Lynn Bannick Kemner, 2007 Objective
Workbook for Simplified ICSE Chemistry, *Stoichiometry Unit Project* Luann Marie Decker, 1998 **General Chemistry**
Workbook Daniel C. Tofan, 2010-07-28 This workbook is a comprehensive collection of solved exercises and problems typical to AP introductory and general chemistry courses as well as blank worksheets containing further practice problems and questions It contains a total of 197 learning objectives grouped in 28 lessons and covering the vast majority of the types of problems that a student will encounter in a typical one year chemistry course It also contains a fully solved 50 question practice test which gives students a good idea of what they might expect on an actual final exam covering the entire material

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