



Chemistry Chapter 10

Name _____

Stoich Worksheet 2

Stoichiometry Practice Problems 1

Do this on a separate sheet of paper. **Show all work and use units.** Round your answers to 3 significant digits.

1. What mass of iron is needed to react completely with 64.2 g of sulfur?



2. How much copper (I) oxide is needed to produce 75.0 g of copper (I) chloride?



3. How much hydrogen sulfite is produced when 178 g of sulfur dioxide reacts with water?



4. If 65.0 g of iron (II) sulfide is added to hydrogen chloride, what mass of hydrogen sulfide is produced?



5. Silver bromide is a precipitate that is formed when silver nitrate reacts with sodium bromide. What mass of the precipitate is produced if you start with 48.7 g of silver nitrate?



6. How much hydrogen nitrate is needed to react completely with 36.3 g of magnesium?



7. What mass of hydrogen can be produced when 82.1 g of zinc reacts with hydrogen sulfate?



For #8 & 9, you will need to use the density of oxygen gas is 1.43g/L.

8. What volume of oxygen is produced when 28.9g of water is decomposed?



9. What volume of oxygen is produced from the decomposition of 145.7g of potassium chlorate?



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Robert R. Redfield



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Table of Contents Stoichiometry Problems Word

1. Understanding the eBook Stoichiometry Problems Word
 - The Rise of Digital Reading Stoichiometry Problems Word
 - Advantages of eBooks Over Traditional Books
2. Identifying Stoichiometry Problems Word
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Stoichiometry Problems Word
 - User-Friendly Interface
4. Exploring eBook Recommendations from Stoichiometry Problems Word
 - Personalized Recommendations
 - Stoichiometry Problems Word User Reviews and Ratings
 - Stoichiometry Problems Word and Bestseller Lists
5. Accessing Stoichiometry Problems Word Free and Paid eBooks
 - Stoichiometry Problems Word Public Domain eBooks
 - Stoichiometry Problems Word eBook Subscription Services

- Stoichiometry Problems Word Budget-Friendly Options
- 6. Navigating Stoichiometry Problems Word eBook Formats
 - ePub, PDF, MOBI, and More
 - Stoichiometry Problems Word Compatibility with Devices
 - Stoichiometry Problems Word Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Stoichiometry Problems Word
 - Highlighting and Note-Taking Stoichiometry Problems Word
 - Interactive Elements Stoichiometry Problems Word
- 8. Staying Engaged with Stoichiometry Problems Word
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Stoichiometry Problems Word
- 9. Balancing eBooks and Physical Books Stoichiometry Problems Word
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Stoichiometry Problems Word
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Stoichiometry Problems Word
 - Setting Reading Goals Stoichiometry Problems Word
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Stoichiometry Problems Word
 - Fact-Checking eBook Content of Stoichiometry Problems Word
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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