

Section 1

SECTION 1

Answer each question by circling the correct answer(s) in the space provided.

1. _____ The following is a chemical equation representing the reaction of hydrogen gas with oxygen gas to form water.
 - (a) correct, balanced, all atoms and products
 - (b) correct, balanced, all atoms and products
 - (c) correct, all atoms of water formed in water, all atoms of oxygen formed in oxygen
 - (d) correct, all atoms of water formed in water, all atoms of oxygen formed in oxygen
2. _____ The following is a chemical equation representing the reaction of hydrogen gas with oxygen gas to form water.
 - (a) the mass of water formed is 18 g
 - (b) the mass of water formed is 18 g
 - (c) the mass of water formed is 18 g
 - (d) the mass of water formed is 18 g
3. _____ The following is a chemical equation representing the reaction of hydrogen gas with oxygen gas to form water.
 - (a) correct, balanced, all atoms and products
 - (b) correct, balanced, all atoms and products
 - (c) correct, balanced, all atoms and products
 - (d) correct, balanced, all atoms and products
4. _____ The following is a chemical equation representing the reaction of hydrogen gas with oxygen gas to form water.
 - (a) correct, balanced, all atoms and products
 - (b) correct, balanced, all atoms and products
 - (c) correct, balanced, all atoms and products
 - (d) correct, balanced, all atoms and products

5. Given the following equation, the following is a chemical equation representing the reaction of hydrogen gas with oxygen gas to form water.



a. Balance the equation.



b. Given the following equation, the following is a chemical equation representing the reaction of hydrogen gas with oxygen gas to form water.

1. The mass of water formed is 18 g. The mass of oxygen gas formed is 18 g. The mass of hydrogen gas formed is 18 g.

2. The mass of water formed is 18 g. The mass of oxygen gas formed is 18 g. The mass of hydrogen gas formed is 18 g.

3. The mass of water formed is 18 g. The mass of oxygen gas formed is 18 g. The mass of hydrogen gas formed is 18 g.

Stoichiometry Section Review Answers 9 4

Susanne Siebentritt, Uwe Rau



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