

# Observing and describing chemical reactions

## Aim

The aim of this experiment is to perform simple chemical reactions, whilst handling chemical reagents and simple laboratory glassware so that one may make careful observation and record observations. To allow one to write simple chemical reactions to support the observations made.

## Method

The procedure described in the CEM1008F Practical Manual Experiment 3 (p.3- 6) was followed, excluding steps 2 and 7.

## Results

A table showing the Reactants, Observations and balanced chemical reaction of each experiment.

Experiment	Reactants	Observations	Balanced Chemical equation
1	$\text{AgNO}_3(\text{aq})$ [Silver Nitrate Solution], $\text{Cu}(\text{s})$ [Copper]	Solid Silver began to form on the copper's surface as the silver ions reduced gaining electrons from the oxidised copper. The oxidised copper formed Copper (II) Nitrate making the solution blue. Solid Silver becomes a precipitate of the reaction as a result of the reduction.	$\text{Cu}(\text{s}) + 2\text{AgNO}_3(\text{aq}) \rightarrow \text{Cu}(\text{NO}_3)_2(\text{aq}) + 2\text{Ag}(\text{s})$
3	$\text{Pb}(\text{NO}_3)_2(\text{aq})$ [Lead Nitrate Solution], $\text{Zn}(\text{s})$ [Zinc]	The solution went from colourless to pale yellow as a result of the dissociation reaction that occurred as the Zn replaced the Lead hence, the product Zinc (II) Nitrate is formed which can be attributed to the resultant light yellow.	$\text{Zn}(\text{s}) + \text{Pb}(\text{NO}_3)_2(\text{aq}) \rightarrow \text{Zn}(\text{NO}_3)_2(\text{aq}) + \text{Pb}(\text{s})$
4.1)	$\text{FeCl}_3$ [Iron(III) Chloride], $\text{Na}_2\text{SiO}_3(\text{aq})$ [Sodium Silicate Solution]	When Placed into the Sodium Silicate Solution the Orange Iron (III) Chloride underwent a reaction with the solution and formed Iron Silicate which is responsible for the observed Brown Crystals that one observes growing from the base of the test	$2\text{FeCl}_3(\text{s}) + 3\text{Na}_2\text{SiO}_3(\text{aq}) \rightarrow \text{Fe}_2(\text{SiO}_3)_3(\text{s}) + 6\text{NaCl}(\text{aq})$

# Observing Chemical Reactions Review and Reinforce

**Arturo Cuomo**



## **Observing Chemical Reactions Review and Reinforce:**

## **Observing Chemical Reactions Reviewand Reinforce** Book Review: Unveiling the Power of Words

In a global driven by information and connectivity, the ability of words has become more evident than ever. They have the capacity to inspire, provoke, and ignite change. Such is the essence of the book **Observing Chemical Reactions Reviewand Reinforce**, a literary masterpiece that delves deep into the significance of words and their impact on our lives. Compiled by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we shall explore the book is key themes, examine its writing style, and analyze its overall impact on readers.

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### **Table of Contents Observing Chemical Reactions Reviewand Reinforce**

1. Understanding the eBook Observing Chemical Reactions Reviewand Reinforce
  - The Rise of Digital Reading Observing Chemical Reactions Reviewand Reinforce
  - Advantages of eBooks Over Traditional Books
2. Identifying Observing Chemical Reactions Reviewand Reinforce
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Observing Chemical Reactions Reviewand Reinforce
  - User-Friendly Interface
4. Exploring eBook Recommendations from Observing Chemical Reactions Reviewand Reinforce
  - Personalized Recommendations
  - Observing Chemical Reactions Reviewand Reinforce User Reviews and Ratings
  - Observing Chemical Reactions Reviewand Reinforce and Bestseller Lists

5. Accessing Observing Chemical Reactions Reviewand Reinforce Free and Paid eBooks
  - Observing Chemical Reactions Reviewand Reinforce Public Domain eBooks
  - Observing Chemical Reactions Reviewand Reinforce eBook Subscription Services
  - Observing Chemical Reactions Reviewand Reinforce Budget-Friendly Options
6. Navigating Observing Chemical Reactions Reviewand Reinforce eBook Formats
  - ePub, PDF, MOBI, and More
  - Observing Chemical Reactions Reviewand Reinforce Compatibility with Devices
  - Observing Chemical Reactions Reviewand Reinforce Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Observing Chemical Reactions Reviewand Reinforce
  - Highlighting and Note-Taking Observing Chemical Reactions Reviewand Reinforce
  - Interactive Elements Observing Chemical Reactions Reviewand Reinforce
8. Staying Engaged with Observing Chemical Reactions Reviewand Reinforce
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Observing Chemical Reactions Reviewand Reinforce
9. Balancing eBooks and Physical Books Observing Chemical Reactions Reviewand Reinforce
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Observing Chemical Reactions Reviewand Reinforce
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Observing Chemical Reactions Reviewand Reinforce
  - Setting Reading Goals Observing Chemical Reactions Reviewand Reinforce
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Observing Chemical Reactions Reviewand Reinforce
  - Fact-Checking eBook Content of Observing Chemical Reactions Reviewand Reinforce
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

#### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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