1 This question is about reversible reactions.

Sulfur trioxide is produced in a reversible reaction.

The equation for the reaction is:

sulfur dioxide + oxygen => sulfur trioxide

1.1 What is the correct balanced equation for the reaction?

Tick one box.

2SO₂ + O₂ ⇒ SO₃

SO₂ + 20 ⇒ SO₃

2SO₂ + O₂ ⇒ 2SO₃

2SO₂ + 2O₂ ⇒ 2SO₃

[1 mark]

1.2 The reaction of sulfur dioxide and oxygen to form sulfur trioxide is exothermic.

Which of the following statements is true?

Tick one box.

The activation energy for the reaction of sulfur dioxide with oxygen is negative.

Sulfur trioxide is colder than sulfur dioxide and oxygen.

Heating the reaction will cause more sulfur trioxide to form.

The reaction of sulfur trioxide to form sulfur dioxide and oxygen is endothermic.

[1 mark]

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