

Chap 3 Test Practice**Multiple Choice**

Identify the choice that best completes the statement or answers the question.

- _____ 1. Temperature is a measure of the average _____ energy of the particles in the object.
a. thermal
b. kinetic
c. potential
d. chemical
- _____ 2. The heavier a particle, the _____ it moves.
a. slower
b. faster
c. less
d. more
- _____ 3. What determines the speed of the atoms and molecules of a particular substance?
a. size of the atoms and molecules
b. temperature of the substance
c. Both (a) and (b)
d. None of the above
- _____ 4. Which state of matter will hold its shape without a container?
a. solid
b. liquid
c. gas
d. plasma
- _____ 5. The kinetic theory states that the higher the temperature, the faster the
a. particles that make up a substance move.
b. bonds between atoms break down.
c. molecules of gas rush together.
d. lighter particles within a substance clump together.
- _____ 6. The change of a substance from a solid directly to a gas is called
a. condensation.
b. evaporation.
c. melting.
d. sublimation.
- _____ 7. Evaporation refers to the change of state from a
a. liquid to a gas.
b. gas to a liquid.
c. solid to a liquid.
d. liquid to a solid.
- _____ 8. The law of conservation of mass states that mass cannot be
a. burned.
b. changed in form.
c. created or destroyed.
d. heated or cooled.
- _____ 9. Ice floats in water because it is
a. more dense than water.
b. less dense than water.
c. colder than water.
d. warmer than water.
- _____ 10. Buoyant force is the _____ force exerted on an object immersed or floating on a liquid.
a. lateral
b. upward
c. downward
d. tensile
- _____ 11. As the temperature of a fixed amount of gas at constant volume decreases, its pressure
a. decreases.
b. stays the same.
c. increases.
d. None of the above

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SJ Ball

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