

2013 WASSCE CHEMISTRY OBJECTIVE ANSWER

S/N	CORRECT ANSWER	EXPLANATION
1	B	<p>REMEMBER Radiation refers to the emission or transmission of energy in the form of waves or particles through space or a medium. This energy can take various forms, including electromagnetic waves (such as light, radio waves, and X-rays) or subatomic particles (such as alpha particles, beta particles, and neutrons).</p> <p>EXPLANATION The instrument used in detecting the presence of radiation is Geiger-Muller counter. A Geiger-Muller counter is an instrument specifically designed to detect the presence of ionizing radiation, such as alpha, beta, and gamma radiation. It works based on the ionization effect produced by radiation when it interacts with matter.</p> <p>ANSWER: B</p>
2	D	<p>REMEMBER The linear shape of a molecule refers to the arrangement of its atoms in a straight line, with a bond angle of 180 degrees between them. In the case of carbon dioxide (CO₂), it consists of one carbon (C) atom bonded to two oxygen (O) atoms.</p> <p>EXPLANATION Here's an EXPLANATION of why CO₂ has a linear shape:</p> <ol style="list-style-type: none"> Carbon Atom (C): The carbon atom at the centre of the CO₂ molecule forms two sigma (σ) bonds with the two oxygen atoms. These sigma bonds result from the sharing of electrons between the carbon and oxygen atoms. Oxygen Atoms (O): Each oxygen atom forms a sigma bond with the central carbon atom. This means that each oxygen atom shares two pairs of electrons with the carbon atom. Symmetry: Due to the arrangement of the atoms and the double bonds formed between carbon and oxygen, the CO₂ molecule achieves a linear geometry. Both oxygen atoms are on opposite sides of the carbon atom, and the molecule forms a straight line. Bond Angle: The bond angle between the carbon-oxygen-carbon atoms in CO₂ is approximately 180 degrees, which is characteristic of a linear molecular geometry. <p>So, in summary, the CO₂ molecule is linear because of the arrangement of its</p>

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M Lipman



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