

Post-Lab Questions

1. Based on your X-Y graph, which one of Na_2CO_3 or H_2SO_4 is the limiting reactant in each of the seven flasks in the experiments?

Flask#	1	2	3	4	5	6	7
Limiting Reactant							

2. Based on your experimental results, what is the molar ratio of CO_2 (a product) and Na_2CO_3 (a reactant)?

3. If you react 1.205 g of Na_2CO_3 with 1.205 g of H_2SO_4 , how many grams of CO_2 will be produced.

Stoichiometry By Loss Of CO_2 Lab Answers

D Siedentop



Stoichiometry By Loss Of Co2 Lab Answers:

Energy Research Abstracts, 1991 Semiannual with semiannual and annual indexes References to all scientific and technical literature coming from DOE its laboratories energy centers and contractors Includes all works deriving from DOE other related government sponsored information and foreign nonnuclear information Arranged under 39 categories e g Biomedical sciences basic studies Biomedical sciences applied studies Health and safety and Fusion energy Entry gives bibliographical information and abstract Corporate author subject report number indexes Improving Student Comprehension of Stoichiometric Concepts Connie Lynn Bannick Kemner, 2007

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