#### Unit 08 - Stoichiometry - Worksheet 1

Answer the following questions in the space provided at the left.

18 mol O<sub>2</sub> 1. How many moles of oxygen are produced from twelve moles of Potassium chlorate?

2KCLO<sub>3</sub> → 2KCl + 3O<sub>2</sub>

12 mol KClO<sub>3</sub> 3 mol O<sub>2</sub> 2 mol KClO<sub>3</sub>

21 mol O<sub>2</sub> 2. How many moles of oxygen are produced from 14 moles of Potassium chlorate?

2KCLO<sub>3</sub> → 2KCl + 3O<sub>2</sub>

14 mol KetO<sub>3</sub> 3 mol O<sub>2</sub> 2 mol KClO<sub>3</sub>

\_\_45 mol H<sub>2</sub>O\_\_\_ 3. How many moles of water are produced from 22.5 moles of oxygen?
2H<sub>2</sub> + O<sub>1</sub>Z → 2H<sub>2</sub>O

22.5 mol-O<sub>2</sub> 2 mol H<sub>2</sub>O 1 mol-O<sub>2</sub>

\_20 mol HCI\_\_\_\_ 4. Calculate the number of moles of Hydrogen chloride produced from 10 moles of hydrogen.

H<sub>2</sub> + Cl<sub>2</sub> → 2HCl

10 mol H<sub>2</sub> 2 mol HCl 1 mol H<sub>2</sub>

21 mol Cl<sub>2</sub>
5. Calculate the number of moles of chlorine needed to form 14 moles of Iron (III) chloride.

2Fe + 3Cl<sub>2</sub> → 2FeCl<sub>3</sub>

14 mol EeCl<sub>3</sub> 3 mol Cl<sub>2</sub> 2 mol FeCl<sub>3</sub>

# **Stoichiometry Workbook Answer Key**

**S Ben Porath** 

Stoichiometry Workbook Answer Key:

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In a global used by monitors and the ceaseless chatter of instant communication, the melodic elegance and mental symphony developed by the prepared word usually disappear into the back ground, eclipsed by the relentless noise and disruptions that permeate our lives. But, nestled within the pages of **Stoichiometry Workbook Answer Key** an enchanting fictional prize overflowing with natural thoughts, lies an immersive symphony waiting to be embraced. Crafted by a wonderful musician of language, that captivating masterpiece conducts viewers on a psychological trip, well unraveling the concealed melodies and profound affect resonating within each carefully crafted phrase. Within the depths of the moving review, we shall discover the book is key harmonies, analyze its enthralling publishing fashion, and submit ourselves to the profound resonance that echoes in the depths of readers souls.

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