

12

STOICHIOMETRY

Practice Problems

In your notebook, solve the following problems.

SECTION 12.1 THE ARITHMETIC OF EQUATIONS

Use the 3-step problem-solving approach you learned in Chapter 1.

1. An apple pie needs 10 large apples, 2 cups (top and bottom), and 1 tablespoon of cinnamon. Write a balanced equation that fits this situation. How many apples are needed to make 20 pies?
2. Two moles of potassium chloride and three moles of oxygen are produced from the decomposition of two moles of potassium chlorate, $2\text{KClO}_3(\text{s})$. Write the balanced equation. How many moles of oxygen are produced from 12 moles of potassium chlorate?
3. Using the equation from problem 2, how many moles of oxygen are produced from 14 moles of potassium chlorate?
4. Two molecules of hydrogen react with one molecule of oxygen to produce two molecules of water. How many molecules of water are produced from 3.8×10^{23} molecules of oxygen? How many moles of water are produced from 22.5 moles of oxygen?

SECTION 12.2 CHEMICAL CALCULATIONS

1. Calculate the number of moles of hydrogen chloride produced from 10 moles of hydrogen.



2. Calculate the number of moles of chlorine needed to form 14 moles of iron(II) chloride.



3. Calculate the number of grams of nitrogen dioxide that are produced from 4 moles of nitric oxide.



4. Calculate the mass of oxygen produced from the decomposition of 75.0 g of potassium chlorate.



5. Calculate the mass of silver needed to react with chlorine to produce 94 g of silver chloride. Write a balanced equation first.

6. How many liters of carbon monoxide at STP are needed to react with 4.00 g of oxygen gas to produce carbon dioxide?



7. Calculate the number of liters of oxygen gas needed to produce 15.0 liters of dinitrogen monoxide. Assume all gases are at the same conditions of temperature and pressure.



Stoichiometry 12 Study Guide Answers

Daniel C. Tofan



Stoichiometry 12 Study Guide Answers:

Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual Leo J. Malone, Theodore O. Dolter, 2012-01-03 The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment. New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections, assessment exercises at the end of each section and relevant chapter problems at the end of each chapter. A new Math Check allows quick access to the needed basic skill. The first chapter now includes brief introductions to several fundamental chemical concepts and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem. Every concept in the text is clearly illustrated with one or more step-by-step examples. Making it Real essays have been updated to present timely and engaging real-world applications emphasizing the relevance of the material they are learning. This edition continues the end-of-chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter.

Study Guide to Accompany Calculus for the Management, Life, and Social Sciences Clyde Metz, 1984-01-01 Study Guide to Accompany Calculus for the Management Life and Social Sciences

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Table of Contents Stoichiometry 12 Study Guide Answers

1. Understanding the eBook Stoichiometry 12 Study Guide Answers
 - The Rise of Digital Reading Stoichiometry 12 Study Guide Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Stoichiometry 12 Study Guide Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Stoichiometry 12 Study Guide Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Stoichiometry 12 Study Guide Answers
 - Personalized Recommendations
 - Stoichiometry 12 Study Guide Answers User Reviews and Ratings
 - Stoichiometry 12 Study Guide Answers and Bestseller Lists

5. Accessing Stoichiometry 12 Study Guide Answers Free and Paid eBooks
 - Stoichiometry 12 Study Guide Answers Public Domain eBooks
 - Stoichiometry 12 Study Guide Answers eBook Subscription Services
 - Stoichiometry 12 Study Guide Answers Budget-Friendly Options
6. Navigating Stoichiometry 12 Study Guide Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Stoichiometry 12 Study Guide Answers Compatibility with Devices
 - Stoichiometry 12 Study Guide Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Stoichiometry 12 Study Guide Answers
 - Highlighting and Note-Taking Stoichiometry 12 Study Guide Answers
 - Interactive Elements Stoichiometry 12 Study Guide Answers
8. Staying Engaged with Stoichiometry 12 Study Guide Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Stoichiometry 12 Study Guide Answers
9. Balancing eBooks and Physical Books Stoichiometry 12 Study Guide Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Stoichiometry 12 Study Guide Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Stoichiometry 12 Study Guide Answers
 - Setting Reading Goals Stoichiometry 12 Study Guide Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Stoichiometry 12 Study Guide Answers
 - Fact-Checking eBook Content of Stoichiometry 12 Study Guide Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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