

## Chapter Test B

## A. Matching

Match each item in Column A with the correct description in Column B. Write the letter of the correct description in the blank.

## Column A

- \_\_\_\_\_ 1. An unknown mass dissolved in a known volume of solvent for solution density, a molar.
- \_\_\_\_\_ 2. An unknown volume of gas produced from a known mass of reactant in a closed container at constant.
- \_\_\_\_\_ 3. The relationship of quantities in chemical equations.
- \_\_\_\_\_ 4. The ratio of the actual yield to the theoretical yield, expressed as a percent.
- \_\_\_\_\_ 5. The unknown mass of a product or reactant in a chemical reaction, using a balanced equation.
- \_\_\_\_\_ 6. An unknown volume of a gas produced from a known mass of reactant in a closed container at constant.

## Column B

- a. percent yield
- b. limiting reagent
- c. theoretical yield
- d. stoichiometry
- e. percent yield
- f. percent reagent

## B. Multiple Choice

Choose the best answer for each question from the list.

- \_\_\_\_\_ 1. A 10.0-gram sample of a substance contains 2.0 grams of element A and 8.0 grams of element B. The substance is a compound.  
a. element  
b. mixture  
c. element and compound  
d. none of these
- \_\_\_\_\_ 2. A 10.0-gram sample of a substance contains 2.0 grams of element A and 8.0 grams of element B. The substance is a compound.  
a. 2.0 grams A + 8.0 grams B  
b. 2.0 grams A + 8.0 grams B  
c. 2.0 grams A + 8.0 grams B  
d. 2.0 grams A + 8.0 grams B  
e. 2.0 grams A + 8.0 grams B  
f. 2.0 grams A + 8.0 grams B
- \_\_\_\_\_ 3. A 10.0-gram sample of a substance contains 2.0 grams of element A and 8.0 grams of element B. The substance is a compound.  
a. 2.0 grams A + 8.0 grams B  
b. 2.0 grams A + 8.0 grams B  
c. 2.0 grams A + 8.0 grams B  
d. 2.0 grams A + 8.0 grams B  
e. 2.0 grams A + 8.0 grams B  
f. 2.0 grams A + 8.0 grams B

# Stoichiometry Ch 12 Test

**J Rink**



## **Stoichiometry Ch 12 Test:**

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