STOICHIOMETRY: MIXED PROBLEMS

Name _____

1. $N_2 + 3H_2 \rightarrow 2NH_3$

What volume of NH, at STP is produced if 25.0 g of $\rm N_2$ is reacted with an excess of $\rm H_2?$

2. 2KCIO₃ → 2KCI + 3O₉

If 5.0 g of KCIO, is decomposed, what volume of $\mathrm{O_2}$ is produced at STP?

- 3. How many grams of KCI are produced in Problem 2?
- 4. Zn + 2HCl → ZnCl₂ + H₂

What volume of hydrogen at STP is produced when 2.5 g of zinc react with an excess of hydrochloric acid?

5. H,SO₄ + 2NaOH → H₂O + Na₃SO₄

How many molecules of water are produced if 2.0 g of sodium sulfate are produced in the above reaction?

6. 2AICI, → 2AI + 3CI,

If 10.0 g of aluminum chloride are decomposed, how many molecules of Cl₂ are produced?

Stoichiometry Chemistry If8766 Pg 62

Patrick Vollmar

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This book delves into Stoichiometry Chemistry If8766 Pg 62. Stoichiometry Chemistry If8766 Pg 62 is an essential topic that needs to be grasped by everyone, from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Stoichiometry Chemistry If8766 Pg 62, encompassing both the fundamentals and more intricate discussions.

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 - Chapter 1: Introduction to Stoichiometry Chemistry If8766 Pg 62
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 - ∘ Chapter 3: Stoichiometry Chemistry If8766 Pg 62 in Everyday Life
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