STOICHIOMETRY QUIZ

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4 11

These questions are based on the following unbalanced chemical equation. $C_6H_{18(g)} + O_{2(g)} \longrightarrow CO_{2(g)} + H_2O_{(g)}$ 1a. Write the balanced chemical equation? (2K)

1b. If you were given 1 mol of C_0H_{18} and 10 mols of O_2 , which reactant is limiting? (2K)

1c. Calculate the mass of oxygen needed to produce 4.5×10^{24} molecules of carbon dioxide. (4T)

2a. If 1 g of C₆H₁₈ is combined with 5 g of oxygen, what is the mass of water vapour produced? (4T)

2b. When Dr. Mathewson did this reaction in question 2a he obtained 0.45 g of water. What was the percentage yield? (3T)

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being specific can help the employee feel valued because it shows you took the time to notice their hard work sample winning excellence award nominations excerpted brown university - Mar 02 2023

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