

# LIMITING REACTANT STOICHIOMETRY WORKSHEET

Name _____	Section _____	Date _____
<b>Limiting Reactants</b>		
<b>Directions:</b> Determine the limiting reactant in each problem.		
1) In $2\text{NaOH} + \text{MgCl}_2 \rightarrow \text{Mg(OH)}_2 + 2\text{NaCl}$ , 5.818 grams of NaOH react with 2.712 grams of $\text{MgCl}_2$ .		
2) 0.0718 grams of HCl react with 0.0002 grams of $\text{Ca(OH)}_2$ in the reaction $2\text{HCl} + \text{Ca(OH)}_2 \rightarrow 2\text{H}_2\text{O} + \text{CaCl}_2$ .		
3) 0.2000 grams of Nitrogen in the reaction $\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$ combine with 0.2008 grams of Hydrogen.		

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3) 0.2000 grams of Nitrogen in the reaction $\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$ combine with 0.2008 grams of Hydrogen.		
4) In the reaction $2\text{C}_2\text{H}_2 + 5\text{O}_2 \rightarrow 4\text{CO}_2 + 2\text{H}_2\text{O}$ , 10.000 grams of $\text{O}_2$ react with 0.908 grams of $\text{C}_2\text{H}_2$ .		
5) In the reaction $2\text{Al(OH)}_3 + 3\text{H}_2\text{CO}_3 \rightarrow \text{Al}_2(\text{CO}_3)_3 + 6\text{H}_2\text{O}$ , 1.18 grams of $\text{Al(OH)}_3$ and 1.012 grams of $\text{H}_2\text{CO}_3$ combine.		
6) 0.838 grams of sodium and 0.7021 grams of sodium nitrate react in the following process: $10\text{Na} + 2\text{NaNO}_3 \rightarrow 6\text{Na}_2\text{O} + \text{N}_2$ .		
7) 0.0700 grams of $\text{AlBr}_3$ react with 1.201 grams of $\text{H}_2\text{SO}_4$ to produce products in the following reaction: $2\text{AlBr}_3 + 3\text{H}_2\text{SO}_4 \rightarrow \text{Al}_2(\text{SO}_4)_3 + 6\text{HBr}$ .		
8) When $\text{K}_3\text{PO}_4$ and NaOH react in $\text{K}_3\text{PO}_4 + 3\text{NaOH} \rightarrow 3\text{KOH} + \text{Na}_3\text{PO}_4$ , 5.41 grams of NaOH react with 3.88 grams $\text{K}_3\text{PO}_4$ .		
9) In $\text{Pb(NO}_3)_2 + \text{H}_2\text{S} \rightarrow \text{PbS} + 2\text{HNO}_3$ , 10.02 grams of $\text{Pb(NO}_3)_2$ react with 7.98 grams of $\text{H}_2\text{S}$ .		
10) 0.4000 grams of $\text{H}_2\text{SO}_4$ combine with 0.2480 grams of $\text{Ba(OH)}_2$ in the reaction $\text{H}_2\text{SO}_4 + \text{Ba(OH)}_2 \rightarrow \text{BaSO}_4 + 2\text{H}_2\text{O}$ .		

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10 PROBLEMS

WORKED ANSWER KEY

# Stoichiometry Section 12 3 Limiting Reactants Answer Key

**Silberberg**



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Williamson, Christopher Thompson, 2024-01-09 The third edition of Chemistry Core Concepts Blackman et al has been developed by a group of leading chemistry educators for students entering university with little or no background in chemistry Available as a full colour printed textbook with an interactive eBook code this title enables every student to master concepts and succeed in assessment Lecturers are supported with an extensive and easy to use teaching and learning package *Holt Chemistry* Ralph Thomas Myers, 2004

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