

Stoichiometry Study Guide

Balance the following equations. Then use the balanced equation to answer questions #1-5.

1) 10 moles of H_2O are

a) 180 grams

b) 360 grams

c) 540 grams

2) 10 moles of H_2O are

a) 180 grams

b) 360 grams

c) 540 grams

Balance the following equation.

1) 10 moles of H_2O are

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b) 360 grams

c) 540 grams

2) 10 moles of H_2O are

a) 180 grams

b) 360 grams

c) 540 grams

Name _____ Period _____ Date _____

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Fill in the blanks below.

1 mole = _____ grams.

Balance the following equation. Then use the balanced equation to answer the questions #1-5.



1) Find the formula for the following compounds:

a) NaCl to O_2

b) NaCl to NaClO_2

c) O_2 to NaClO_2

2) If the above chemical reaction is done as an experiment using 4.4 moles of each reactant, calculate how much NaClO_2 will be produced?

a) Given 4.4 moles of NaCl, how many moles of NaClO_2 will be produced?

b) Given 4.4 moles of O_2 , how many moles of NaClO_2 will be produced?

c) Define limiting reactant & excess reactant.

d) Which reactant is the limiting reactant? Which reactant is the excess reactant?

3) If the above chemical reaction is done as an experiment using 10 moles of O_2 , calculate the percent yield given that the actual yield of NaClO_2 is 8.8 moles.

a) What is the theoretical yield in grams of NaClO_2 given 10 moles of the O_2 ?

b) Write the formula for Percent Yield.

c) What is the percent yield given that the actual yield of NaClO_2 is 8.8 moles?

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E Staff



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Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual Leo J. Malone,Theodore O. Dolter,2012-01-03 The 9th edition of Malone s *Basic Concepts of Chemistry* provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections assessment exercises at the end each section and relevant chapter problems at the end of each chapter A new Math Check allows quick access to the needed basic skill The first chapter now includes brief introductions to several fundamental chemical concepts and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem Every concept in the text is clearly illustrated with one or more step by step examples Making it Real essays have been updated to present timely and engaging real world applications emphasizing the relevance of the material they are learning This edition continues the end of chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter ["O" Level Study Guide - Chemistry Quite Easily Done](#) , **Barron's**

Science 360: A Complete Study Guide to Chemistry with Online Practice Barron's Educational Series,Mark Kernion,Joseph A. Mascetta,2021-09-07 provides a complete guide to the fundamentals of chemistry Page 4 of cover

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