

NAME: Key DATE: \_\_\_\_\_ PERIOD: \_\_\_\_\_

# GAS STOICHIOMETRY PROBLEMS WORKSHEET 1

1. Given the unbalanced decomposition reaction of baking soda:



42.0 grams of baking soda? What volumes of carbon dioxide and water are produced at STP?

$$42.0 \text{ g NaHCO}_3 \times \frac{1 \text{ mol NaHCO}_3}{84.01 \text{ g NaHCO}_3} \times \frac{1 \text{ mol CO}_2}{2 \text{ mol NaHCO}_3} \times \frac{22.4 \text{ L CO}_2}{1 \text{ mol CO}_2} = \boxed{5.6 \text{ L CO}_2}$$

$$42.0 \text{ g NaHCO}_3 \times \frac{1 \text{ mol NaHCO}_3}{84.01 \text{ g NaHCO}_3} \times \frac{1 \text{ mol H}_2\text{O}}{2 \text{ mol NaHCO}_3} \times \frac{22.4 \text{ L H}_2\text{O}}{1 \text{ mol H}_2\text{O}} = \boxed{5.6 \text{ L H}_2\text{O}}$$

2. The catalytic decomposition of hydrogen peroxide is:



Balance the reaction. How many moles of water and oxygen are produced by the decomposition of 68.0 grams of hydrogen peroxide? How many molecules of water and oxygen are produced? How many grams of each product are formed?

$$68 \text{ g H}_2\text{O}_2 \times \frac{1 \text{ mol H}_2\text{O}_2}{34.01 \text{ g H}_2\text{O}_2} \times \frac{2 \text{ mol H}_2\text{O}}{2 \text{ mol H}_2\text{O}_2} = \boxed{2.00 \text{ mol H}_2\text{O}}$$

$$68 \text{ g H}_2\text{O}_2 \times \frac{1 \text{ mol H}_2\text{O}_2}{34.01 \text{ g H}_2\text{O}_2} \times \frac{1 \text{ mol O}_2}{2 \text{ mol H}_2\text{O}_2} = \boxed{1.00 \text{ mol O}_2}$$

$$68 \text{ g H}_2\text{O}_2 \times \frac{1 \text{ mol H}_2\text{O}_2}{34.01 \text{ g H}_2\text{O}_2} \times \frac{2 \text{ mol H}_2\text{O}}{2 \text{ mol H}_2\text{O}_2} \times \frac{1.0 \times 10^{24} \text{ molecules H}_2\text{O}}{1 \text{ mol H}_2\text{O}} = \boxed{1.2 \times 10^{24} \text{ molecules H}_2\text{O}}$$

$$68 \text{ g H}_2\text{O}_2 \times \frac{1 \text{ mol H}_2\text{O}_2}{34.01 \text{ g H}_2\text{O}_2} \times \frac{1 \text{ mol O}_2}{2 \text{ mol H}_2\text{O}_2} \times \frac{1.0 \times 10^{24} \text{ molecules O}_2}{1 \text{ mol O}_2} = \boxed{6.0 \times 10^{23} \text{ molecules O}_2}$$

$$68 \text{ g H}_2\text{O}_2 \times \frac{1 \text{ mol H}_2\text{O}_2}{34.01 \text{ g H}_2\text{O}_2} \times \frac{2 \text{ mol H}_2\text{O}}{2 \text{ mol H}_2\text{O}_2} \times \frac{18.02 \text{ g H}_2\text{O}}{1 \text{ mol H}_2\text{O}} = \boxed{36.02 \text{ g H}_2\text{O}}$$

$$68 \text{ g H}_2\text{O}_2 \times \frac{1 \text{ mol H}_2\text{O}_2}{34.01 \text{ g H}_2\text{O}_2} \times \frac{1 \text{ mol O}_2}{2 \text{ mol H}_2\text{O}_2} \times \frac{32.00 \text{ g O}_2}{1 \text{ mol O}_2} = \boxed{63.96 \text{ g O}_2}$$

3. If 156.0 grams of potassium metal reacts with excess water, potassium hydroxide and hydrogen gas are formed? What volume of hydrogen gas, in liters,

Be sure to balance the reaction first.



$$156 \text{ g K} \times \frac{1 \text{ mol K}}{39.09 \text{ g K}} \times \frac{1 \text{ mol H}_2}{2 \text{ mol K}} \times \frac{22.4 \text{ L H}_2}{1 \text{ mol H}_2} = \boxed{44.70 \text{ L H}_2}$$

$$156 \text{ g K} \times \frac{1 \text{ mol K}}{39.09 \text{ g K}} \times \frac{2 \text{ mol KOH}}{2 \text{ mol K}} \times \frac{56.11 \text{ g KOH}}{1 \text{ mol KOH}} = \boxed{87.37 \text{ g KOH}}$$

4. Determine the number of moles of carbon dioxide gas, water, and sodium chloride formed by the reaction of 42.0 grams of sodium bicarbonate (baking soda) reacting with excess hydrochloric acid. The reaction is:



$$42 \text{ g NaHCO}_3 \times \frac{1 \text{ mol NaHCO}_3}{84.01 \text{ g NaHCO}_3} \times \frac{1 \text{ mol CO}_2}{1 \text{ mol NaHCO}_3} = \boxed{0.5 \text{ mol CO}_2}$$

$$42 \text{ g NaHCO}_3 \times \frac{1 \text{ mol NaHCO}_3}{84.01 \text{ g NaHCO}_3} \times \frac{1 \text{ mol H}_2\text{O}}{1 \text{ mol NaHCO}_3} = \boxed{0.5 \text{ mol H}_2\text{O}}$$

$$42 \text{ g NaHCO}_3 \times \frac{1 \text{ mol NaHCO}_3}{84.01 \text{ g NaHCO}_3} \times \frac{1 \text{ mol NaCl}}{1 \text{ mol NaHCO}_3} = \boxed{0.5 \text{ mol NaCl}}$$

# Stoichiometry Test Answers

**Joseph A. Mascetta, Mark Kernion**

A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the authors' names.

## **Stoichiometry Test Answers:**

**AP Chemistry For Dummies** Peter J. Mikulecky, Michelle Rose Gilman, Kate Brutlag, 2008-11-13 A practical and hands on guide for learning the practical science of AP chemistry and preparing for the AP chem exam Gearing up for the AP Chemistry exam AP Chemistry For Dummies is packed with all the resources and help you need to do your very best Focused on the chemistry concepts and problems the College Board wants you to know this AP Chemistry study guide gives you winning test taking tips multiple choice strategies and topic guidelines as well as great advice on optimizing your study time and hitting the top of your game on test day This user friendly guide helps you prepare without perspiration by developing a pre test plan organizing your study time and getting the most out of your AP course You ll get help understanding atomic structure and bonding grasping atomic geometry understanding how colliding particles produce states and so much more To provide students with hands on experience AP chemistry courses include extensive labwork as part of the standard curriculum This is why the book dedicates a chapter to providing a brief review of common laboratory equipment and techniques and another to a complete survey of recommended AP chemistry experiments Two full length practice exams help you build your confidence get comfortable with test formats identify your strengths and weaknesses and focus your studies You ll discover how to Create and follow a pretest plan Understand everything you must know about the exam Develop a multiple choice strategy Figure out displacement combustion and acid base reactions Get familiar with stoichiometry Describe patterns and predict properties Get a handle on organic chemistry nomenclature Know your way around laboratory concepts tasks equipment and safety Analyze laboratory data Use practice exams to maximize your score Additionally you ll have a chance to brush up on the math skills that will help you on the exam learn the critical types of chemistry problems and become familiar with the annoying exceptions to chemistry rules Get your own copy of AP Chemistry For Dummies to build your confidence and test taking know how so you can ace that exam

**MCAT Exam Prep** Bill T Reese, 2024-05-13 What does it take to embark on the journey towards a career in medicine Dive into the world of aspiring medical professionals with our comprehensive guide designed to equip you with the knowledge skills and strategies needed to conquer the Medical College Admission Test MCAT and pave the way for success in medical school We unravel the mysteries of the MCAT through a series of engaging chapters that delve into the core subjects tested on the exam From biology and biochemistry to physics and organic chemistry each chapter offers a deep exploration of essential concepts accompanied by practice questions and detailed explanations to reinforce your understanding But mastering the MCAT is not just about memorizing facts and formulas it s about honing your critical thinking analytical reasoning and problem solving abilities That s why our book goes beyond mere content review to provide valuable insights into effective study strategies test taking techniques and mental preparation tips to help you perform at your best on exam day Whether you re a pre med student embarking on your MCAT journey or a seasoned test taker seeking to improve your scores this book is your trusted companion every step of the way

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*Roadmap to the Regents* Sasha Alcott, 2003 If Students Need to Know It It's in This Book This book develops the chemistry skills of high school students. It builds skills that will help them succeed in school and on the New York Regents Exams. Why The Princeton Review We have more than twenty years of experience helping students master the skills needed to excel on standardized tests. Each year we help more than 2 million students score higher and earn better grades. We Know the New York Regents Exams Our experts at The Princeton Review have analyzed the New York Regents Exams and this book provides the most up to date thoroughly researched practice possible. We break down the test into individual skills to familiarize students with the test's structure while increasing their overall skill level. We Get Results We know what it takes to succeed in the classroom and on tests. This book includes strategies that are proven to improve student performance. We provide a breakdown of the skills based on New York standards and objectives, hundreds of practice questions organized by skill, two complete practice New York Regents Exams in Physical Setting Chemistry.

SAT Subject Test Chemistry Joseph A. Mascetta, Mark Kernion, 2020-12-01 Note: College Board has discontinued the SAT Subject Tests in the US. The tests will be available outside the US in June 2021 and then be discontinued. Barron's SAT Subject Test Chemistry with 7 Practice Tests features in-depth review of all topics on the exam and full-length practice tests in the book and online. This edition includes One full-length diagnostic test to help you assess your strengths and weaknesses. Comprehensive review of all topics on the exam including introductory chemistry, atomic structure and the periodic table, bonding, chemical formulas, gases and laws, stoichiometry, liquids, solids, and phase changes, chemical reactions and thermochemistry, chemical reactions, chemical equilibrium, acids, bases, and salts, oxidation-reduction, carbon and organic chemistry, and the laboratory. Four full-length practice tests that reflect the actual SAT Subject Test Chemistry exam in length, question types, and degree of difficulty. Two full-length online practice tests with answer explanations and automated scoring. Appendices which include the periodic table, important equation, constant, and data tables, and a glossary of chemistry terms.

**McGraw-Hill's SAT Subject Test Chemistry, 3rd Edition** Thomas Evangelist, 2011-02-03 Expert guidance on the Chemistry exam. Many colleges and universities require you to take one or more SAT II Subject Tests to demonstrate your mastery of specific high school subjects. McGraw-Hill's SAT Subject Test Chemistry is written by experts in the field and gives you the guidance you need to perform at your best. This book includes 4 full-length sample tests updated for the latest test format, 40 top tips to remember on test day, Glossary of tested chemistry terms and formulas, Tips and strategies from one of the most popular teachers at the renowned Brooklyn Technical High School, Diagnostic test to pinpoint strengths and weaknesses, Step-by-step review of all topics covered on the exam, In-depth coverage of the lab experiment questions that are a major test feature, Charts, tables, and illustrations to simplify and reinforce learning, Practice tests just like the real SAT Subject Test in Chemistry, Test-taking tips and strategies.

**Cracking**

**the AP Chemistry** Paul Foglino, Princeton Review (Firm), 2004 The fiercer the competition to get into college the more schools require that students prove themselves in other ways than SAT scores and grade point averages The more expensive college educations become the more students take advantage of the opportunity to test out off first year college courses Includes 2 sample tests with full explanations for all answers The Princeton Review's proven score raising skills and techniques Complete subject review of all the material likely to show up on the AP Chemistry exam **Intelligent Tutoring Systems** Stefan Trausan-Matu, Kristy Boyer, Martha Crosby, Kitty Panourgia, 2014-05-24 This book constitutes the refereed proceedings of the 12th International Conference on Intelligent Tutoring Systems ITS 2014 held in Honolulu HI USA in June 2014 The 31 revised full papers 45 short papers and 27 posters presented were carefully viewed and selected from 177 submissions The specific theme of the ITS 2014 conference is Creating fertile soil for learning interactions Besides that the highly interdisciplinary ITS conferences bring together researchers in computer science learning sciences cognitive and educational psychology sociology cognitive science artificial intelligence machine learning and linguistics The papers are organized in topical sections on affect multimodality and metacognition collaborative learning data mining and student behavior dialogue and discourse generating hints scaffolds and questions game based learning and simulation graphical representations and learning student strategies and problem solving scaling ITS and assessment **McGraw-Hill's SAT Subject Test: Chemistry, 2ed** Thomas Evangelist, 2009-02-01 We want to help you score high on the SAT Chemistry test We've put all of our proven expertise into McGraw Hill's SAT Subject Test Chemistry to make sure you're fully prepared for this difficult exam With this book you'll get essential skill building techniques and strategies created by a leading high school chemistry teacher You'll also get 5 full length practice tests hundreds of sample questions and all the facts about the current exam With McGraw Hill's SAT Subject Test Chemistry we'll guide you step by step through your preparation program and give you the tools you need to succeed 5 full length sample tests with complete explanations for every question 40 top test items to remember on exam day A step by step review of all topics covered on the exam Teacher recommended tips and strategies to help you raise your score [The Official ACT Prep Pack 2019-2020 with 7 Full Practice Tests, \(5 in Official ACT Prep Guide + 2 Online\)](#) ACT, 2019-05-07 Get The Official ACT Prep Guide 2019 2020 plus hours of ACT Online Prep with this unique bundle from the makers of the ACT This updated edition includes The Official ACT Prep Guide 2019 2020 Six months access to the ACT Online Prep platform Seven full length practice tests including a NEW never before seen test five tests in book and two online Optional writing tests More than 2 400 online questions that can be filtered and organized into practice sets Detailed explanations for every answer along with tips for boosting your score on the English math reading science and optional writing tests Free mobile app for on the go learning Game center to test further knowledge Flashcards customized for individual review needs and learning paths Daily goals and tracking to help maintain focus Advice and guidance for test day The Official ACT Prep Pack 2019 2020 is the only guide from the makers of the ACT exam and includes actual ACT test

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