

Stoichiometry Worksheet 2: Percent Yield

For each of the problems below:

- Write the balanced chemical equation
- Identify the given (with units) and what you want to find (with units)
- Show set up with units. Check sig figs, give final answer with units and label.

1. Using the Hoffman apparatus for electrolysis, a chemist decomposes 36 g of water into its gaseous elements. How many grams of hydrogen gas should she get (theoretical yield)?



Before $\underline{2 \text{ mol}}$ $\underline{0 \text{ mol}}$ $\underline{0 \text{ mol}}$

Change $\underline{-2 \text{ mol}}$ $\underline{+2 \text{ mol}}$ _____

After $\underline{0 \text{ mol}}$ $\underline{2 \text{ mol}}$

$$36 \text{ g H}_2\text{O} \times \frac{1 \text{ mol H}_2\text{O}}{18.0 \text{ g H}_2\text{O}} = 2.0 \text{ mol H}_2\text{O} \times \frac{2 \text{ mol H}_2}{2 \text{ mol H}_2\text{O}} = 2 \text{ mol H}_2 \times \frac{2.0 \text{ g H}_2}{1 \text{ mol H}_2} = 4.0 \text{ g H}_2$$

2. Recall that liquid sodium reacts with chlorine gas to produce sodium chloride. You want to produce 581 g of sodium chloride. How many grams of sodium are needed?



Before $\underline{9.94 \text{ mol}}$ $\underline{\text{xs mol}}$ $\underline{0 \text{ mol}}$

Change $\underline{-9.94 \text{ mol}}$ _____ $\underline{+9.94 \text{ mol}}$

After $\underline{0 \text{ mol}}$ $\underline{9.94 \text{ mol}}$

$$581 \text{ g NaCl} \times \frac{1 \text{ mol NaCl}}{58.5 \text{ g NaCl}} = 9.93 \text{ mol NaCl} \times \frac{2 \text{ mol Na}}{2 \text{ mol NaCl}} = 9.93 \text{ mol Na} \times \frac{23.0 \text{ g Na}}{1 \text{ mol Na}} = 228 \text{ g Na}$$

3. You eat 180.0 g of glucose (90 M&Ms). If glucose, $\text{C}_6\text{H}_{12}\text{O}_6$, reacts with oxygen gas to produce carbon dioxide and water, how many grams of oxygen will you have to breathe in to burn the glucose?



Before $\underline{1.0 \text{ mol}}$ $\underline{\text{xs mol}}$ $\underline{0 \text{ mol}}$ $\underline{0 \text{ mol}}$

Change $\underline{-1.0 \text{ mol}}$ $\underline{-6.0 \text{ mol}}$ _____

After $\underline{0 \text{ mol}}$ $\underline{\text{xs mol}}$

$$180.0 \text{ g C}_6\text{H}_{12}\text{O}_6 \times \frac{1 \text{ mol C}_6\text{H}_{12}\text{O}_6}{180 \text{ g C}_6\text{H}_{12}\text{O}_6} = 1.00 \text{ mol C}_6\text{H}_{12}\text{O}_6 \times \frac{6 \text{ mol O}_2}{1 \text{ mol C}_6\text{H}_{12}\text{O}_6} = 6.00 \text{ mol O}_2$$

$$6.0 \text{ mol O}_2 \times \frac{32.0 \text{ g O}_2}{1 \text{ mol O}_2} = 192.0 \text{ g O}_2$$

Stoichiometry Answer Key Simulation

Sebastian Brünink

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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Water Quality Modeling Wu-Seng Lung,1993-07-29 This volume to discussing the various aspects of estuarine water quality modeling Topics considered include fundamental principles estuarine mass transport BOD DO and eutrophication model kinetics kinetics on toxicants and sediment water interactions The book also discusses mixing zone modeling and how to integrate estuarine hydrodynamic and water quality models Many case studies demonstrating successful model applications are discussed **Prentice Hall Chemistry** ,2000 **Control and Safety Analysis of Intensified Chemical Processes** Dipesh Shikchand Patle,Gade Pandu Rangaiah,2024-03-25 Resource on the control and safety analysis of intensified chemical processes ranging from general methods to specific applications Control and Safety Analysis of Intensified Chemical Processes covers the basic principles of and recent developments in control and safety analysis of intensified chemical processes ranging from dynamic simulations and safety analysis to the design and control of important processes The text discusses general methods and tools such as dynamic simulation control and safety analysis as well as design aspects and analysis of important applications in order to provide scientists and engineers with an understanding of the design control and safety considerations involved in intensified chemical processes Sample topics covered in Control and Safety Analysis of Intensified Chemical Processes include Simulation and optimization methods common programs and simulators for simulation and optimization and interfacing of simulators and optimizers Programs simulators for dynamic simulation and control tuning of controllers and popular criteria for control assessment Control of a hybrid reactive extractive distillation systems for ternary azeotropic mixtures reactive distillation in recycle systems and middle vessel batch distillation with vapor recompression Safety analysis of intensified processes e g extractive distillation dividing wall column dividing wall column with mechanical vapor recompression and algal biodiesel process A comprehensive resource on the subject Control and Safety Analysis of Intensified Chemical Processes is a highly valuable reference for researchers students and practitioners interested in process intensification and their applications The text can be adopted by instructors for use in advanced courses on process control and safety **Research on Education in Africa, the Caribbean, and the Middle East** Kagendo Mutua,Cynthia S Sunal,2009-03-01 **The Software Encyclopedia** ,1988 Fluorine Metabolism, Transport and Enzymatic Chemistry ,2024-04-23 Fluorine Metabolism Transport and Enzymatic Chemistry Volume 696 in the Methods in Enzymology series highlights new advances in the field with this new volume presenting interesting topics on Fluorine Metabolism Transport and Enzymatic Chemistry Chapters in this new release include Determination of fluoride sensitivity and resistance in oral biofilm models NMR methods to detect fluoride binding and transport by membrane proteins Biotransformation of fluorinated drugs and xenobiotics by the model fungus Cunninghamella elegans Measurement of fluoride efflux from lipid vesicles and cells Defluorination of PFAS by Acidimicrobium sp A6 and Potential Applications for Remediation and more Additional chapters cover Computational approaches to investigate F binding selectivity and transport

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Introductory Chemistry Steve Russo, Mike Silver, 2002 Steve Russo and Mike Silver turn chemistry into a memorable story that engages readers and provides the context they need to understand and remember core concepts The book builds interesting applications and well designed illustrations into the narrative to get and hold attention then builds confidence with integrated active learning activities Readers make the connections between concepts and the problem solving techniques they need to master as they read The new edition strengthens this conceptual approach and presents additional quantitative techniques in key areas Readers will find enhanced support for quantitative problem solving and more challenging questions at the end of each chapter in addition to the wealth of technology based support on The Chemistry Place tm Special Edition and on The Chemistry of Life CD ROM For college instructors and students *Resources in education*, 1984-06 **Diffusion** E. L. Cussler, 2009-01-15 The clearest coverage available of diffusion and mass transfer which is a key part of the chemical engineering curriculum *The Software Encyclopedia 2000* Bowker Editorial Staff, 2000-05 Innovative Curriculum Materials, 1999 *Computers in Education*, 1975 *Metacognition in Science Education* Anat Zohar, Yehudit Judy Dori, 2011-10-20 Why is metacognition gaining recognition both in education generally and in science learning in particular What does metacognition contribute to the theory and practice of science learning Metacognition in Science Education discusses emerging topics at the intersection of metacognition with the teaching and learning of science concepts and with higher order thinking more generally The book provides readers with a background on metacognition and analyses the latest developments in the field It also gives an account of best practice methodology Expanding on the theoretical underpinnings of metacognition and written by world leaders in metacognitive research the chapters present cutting edge studies on how various forms of metacognitive instruction enhance understanding and thinking in science classrooms The editors strive for conceptual coherency in the various definitions of metacognition that appear in the book and show that the study of metacognition is not an end in itself Rather it is integral to other important constructs such as self regulation literacy the teaching of thinking strategies motivation meta strategies conceptual understanding reflection and critical thinking The book testifies to a growing recognition of the potential value of metacognition to science learning It will motivate science educators in different educational contexts to incorporate this topic into their ongoing research and practice **ENC Focus**, 2000 **Ocean Ecology** J. Emmett Duffy, 2021-08-10 Marine ecology is more interdisciplinary broader in scope and more intimately linked to human activities than ever before Ocean

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Michael, Steve Russo, 2001-12 Introductory Chemistry Third Edition helps readers master the quantitative skills and conceptual understanding they need to gain a deep understanding of chemistry Unlike other books on the market that emphasize rote memory of problem solving algorithms Introductory Chemistry takes a conceptual approach with the idea that focusing on the concepts behind chemical equations helps readers become more proficient problem solvers What Is Chemistry The Numerical Side of Chemistry The Evolution of Atomic Theory The Modern Model of the Atom 1 Chemical Bonding and Nomenclature The Shape of Molecules Chemical Reactions Stoichiometry and the Mole The Transfer of Electrons from One Atom to Another in a Chemical Reaction Intermolecular Forces and the Phases of Matter What If There Were No Intermolecular Forces The Ideal Gas Solutions When Reactants Turn into Products Chemical Equilibrium Electrolytes Acids and Bases For all readers interested in introductory chemistry **Dynamic Modeling & Expert Sys in Wastewater Engin** Gilles G. Patry, David T. Chapman, 1989-09 Dynamic modeling of suspended growth biological wastewater treatment processes Mathematical modeling of fixed film growth Continuous settler operation a dynamic model Dynamics stability and control of the anaerobic digestion process Wastewater treatment and receiving water body interactions Applications of expert systems in the process industry Knowledge based system for the diagnosis of an activated sludge plant Applications of expert system in environmental engineering System identification and control Practical experiences of identification and modeling from experiments Dynamic modeling and expert systems in wastewater engineering trends problems needs Encyclopedia of Ecology Brian D. Fath, 2014-11-03 The groundbreaking Encyclopedia of Ecology provides an authoritative and comprehensive coverage of the complete field of ecology from general to applied It includes over 500 detailed entries structured to provide the user with complete coverage of the core knowledge accessed as intuitively as possible and heavily cross referenced Written by an international team of leading experts this revolutionary encyclopedia will serve as a one stop shop to concise stand alone articles to be used as a point of entry for undergraduate students or as a tool for active researchers looking for the latest information in the field Entries cover a range of topics including Behavioral Ecology Ecological Processes Ecological Modeling Ecological Engineering Ecological Indicators Ecological Informatics Ecosystems Ecotoxicology Evolutionary Ecology General Ecology Global Ecology Human Ecology System Ecology The first reference work to cover all aspects of ecology from basic to applied Over 500 concise stand alone articles are written by prominent leaders in the field Article text is supported by full color photos drawings tables and other visual material Fully indexed and cross referenced with detailed references for further study Writing level is suited to both the expert and non expert Available electronically on ScienceDirect shortly upon publication

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