Chemistry: Storchromerry

How many moles of 0, should be supplied to:

burn 1 mol of C/n (propane) molecules in a camping stone?

 Hom many moles of Q₁ molecules should has

supplied to burn 1 mol of OK, molecules in a domestic furnace?

 Sodium thiosulfate (MacSuDJ), philosopapher is

Thype" reacts with unexposed silver branide in the film emulsion to form sodium branide and a compound of formula NaúAgiSub). If they many moles of hapting, formula units are needed to make to 0.10 and of high sublible?

- Calculate the mass of alumina (MVD): produced when 200 g of aluminum forms in progets
- "Staked firms." Calibro L is formed from "quick-firms" (Calib by adding water. What make of reater is needed to convert 50 kg of quicklime to staked firms? What make of staked lime is produced?
 - Carmels stone the fat tristearin (CutturGu)

the hump. As mell as being a source of energy, the fat is a source of mater, because when it is used the reaction.

344,000,00

taken place. What mass of mater is available from 2.0 kg of fat?

 This compound discrene DIUHJ mas at one.

time considered for use as a rocket fact. Howemany grams of liquid oxygen would a rocket have to carry to burn 20 kg of diborane conglishing. (The products of the condustion are \$40, and \$40).

Oliven the balanced cherrical equation.

How many motes of sodium bronkde (Nath) could be produced from 9.372 mol of bronkne (BrJ7)

3. How many formula units of calcium oxide

(Ca0) can be produced from 4.9 × 50* molecules of oxygen gas (Ou) that reset with calcium (Ca) according to this balanced chemical equation?

2 Call # 0 (5a) # 2 Ca01 #

20. Aluminum metal (Al) neacts with sulfur (S)

produce aluminum suffide (MySu) according to this balanced charms at equation

LINETHING REASONITE

31. What is the maximum mass of methane

that can be burned if only 2.0 g of oxygen is available?

- What is the maximum mass of gluciuse (CHL-DL) that can be burned in 10 g of snygen?
- The solid fixed in the booster stage of the space shuttle is a invistore of ammonium parchlorate.

and aluminum powider, which react as follows:

HARRISON AND

What mass of aluminum should be reced with 5.0 × 20° kg of ammortum perchlorate. If the reaction as stateoff

34. A solution containing 5.0 g of silver nitrate reas mixed with another containing 5.0 g of notices.

chloride. Which was the limiting reagent for the precipitation of silver chloride?

15. Given the balanced charried equation.

How many atoms of silver metal (Agi) are required to react completely with \$35.5 g of loding (I) to produce silver loding (Agi)?

36. The the cretical yield of anymonia in an industrial synthesis was 550 time, but only 460 time was

obtained. What was the percentage yield of the

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Matt Kingsley

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