

Stoichiometry Worksheet 2

1. Sodium chloride can be produced by the reaction of sodium and chlorine gas. Write a balanced equation for this reaction.

Suppose that 6.70 mol Na react with 3.20 mol Cl_2 .

- What is the limiting reagent?
- How many moles of NaCl are produced?

2. The equation for the complete combustion of ethane, C_2H_6 , is



Balance the equation. If 2.70 mol C_2H_6 is reacted with 6.30 mol O_2 ,

- identify the limiting reagent.
- calculate the moles of water produced.

3. Copper and sulfur combine to form copper (I) sulfide.



- What is the limiting reagent when 80.0 g Cu reacts with 25.0 g S?
- What is the maximum number of grams of Cu_2S that can be formed?

Stoichiometry Supplemental Problems Answers

Michael B. Cutlip, Mordechai Shacham



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Cooperative Chemistry Lab Manual Cooper,2005-02 The laboratory course described in the lab manual emphasizes experimental design data analysis and problem solving Inherent in the design is the emphasis on communication skills both written and oral Students work in groups on open ended projects in which they are given an initial scenario and then asked to investigate a problem There are no formalized instructions and students must plan and carry out their own investigations

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