

Pre-Lab Assignment: Stoichiometry Lab

1. In a lab experiment similar to the one you are going to do, a student slowly added hydrochloric acid (HCl) to an evaporating dish containing magnesium carbonate MgCO_3 .
- a. **Complete** and **balance** the following equation for the reaction between magnesium carbonate and hydrochloric acid.



(When writing the formula for the missing product did you consider the charges of the ions?)

- b. When carbonic acid is formed in a chemical reaction, it is unstable and immediately decomposes into carbon dioxide and water. Rewrite the equation above to show the final products using this information.
- c. If the products are heated to dryness which one would be left in the evaporating dish?
- d. To calculate the amount of product theoretically produced in this reaction, the first step is to convert the grams of reactant into moles. Practice this step by converting by converting 2.050 g of MgCO_3 into moles. Show your calculations using dimensional analysis.

Stoichiometry Lab Pre Lab Answers

Javier García-Martínez, Elena Serrano-Torregrosa



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Stoichiometry Unit Project Luann Marie Decker,1998 Instructor's Guide for Introductory Chemistry in the Laboratory James F. Hall,1996 *Laboratory Experiments for General Chemistry* Harold R. Hunt,Toby F. Block,George M. McKelvy,2002 This established manual focuses on using non hazardous materials to teach the experimental nature of general chemistry Experiments are written to address students of various academic backgrounds and differing interests and abilities in chemistry While most experiments can be conducted in a single three hour period some have been designed to be completed over an extended time to illustrate that chemical systems do not work at an arbitrary schedule Suggestions are provided for combining experiments of shorter length and similar pedagogy The Organic Chem Lab Survival Manual James W. Zubrick,2020-02-05 Teaches students the basic techniques and equipment of the organic chemistry lab the updated new edition of the popular hands on guide The Organic Chem Lab Survival Manual helps students understand the basic techniques essential safety protocols and the standard instrumentation necessary for success in the laboratory Author James W Zubrick has been assisting students navigate organic chemistry labs for more than three decades explaining how to set up the laboratory make accurate measurements and perform safe and meaningful experiments This practical guide covers every essential area of lab knowledge from keeping detailed notes and interpreting handbooks to using equipment for chromatography and infrared spectroscopy Now in its eleventh edition this guide has been thoroughly updated to cover current laboratory practices instruments and techniques Focusing primarily on macroscale equipment and experiments chapters cover microscale jointware drying agents recrystallization distillation nuclear magnetic resonance and much more This popular textbook Familiarizes students with common lab instruments Provides guidance on basic lab skills and procedures Includes easy to follow diagrams and illustrations of lab experiments Features practical exercises and activities at the end of each chapter Provides real world examples of lab notes and instrument manuals The Organic Chem Lab Survival Manual A Student s Guide to Techniques 11th Edition is an essential resource for students new to the laboratory environment as well as those more experienced seeking to refresh their knowledge *Using Multimedia Technology in Chemistry Pre-laboratory Preparation* Jeffrey Glen Yoder,2002 **Improving Student Comprehension of Stoichiometric Concepts** Connie Lynn Bannick Kemner,2007 **A Concrete Stoichiometry Unit for High School Chemistry** Jennifer Louise Pakkala,2006 **General Chemistry** B. Richard Siebring,Mary Ellen Schaff,1980 Optimizing STEM Education With Advanced ICTs and Simulations Levin, Ilya,Tsybulsky, Dina,2017-06-05 The role of technology in educational settings has become increasingly prominent in recent years When utilized effectively these tools provide a higher quality of learning for students Optimizing STEM Education With Advanced ICTs and Simulations is an innovative reference source for the latest scholarly research on the integration of digital tools for enhanced STEM based learning environments Highlighting a range of pivotal topics such as mobile games virtual labs and participatory simulations this publication is ideally designed for

educators professionals academics and students seeking material on emerging educational technologies

Reduction

Roasting-acid Solution Techniques in Laboratory Processing of Minnesota Manganiferous Ores Perry L. Weston, M. M. Fine, 1966 The Bureau of Mines used high temperature reduction roasting magnetic separation acid leaching and autoclave precipitation processes to recover iron and manganese from brown Cuyana range ores of Minnesota Wet magnetic separation concentrated over 90 percent of the iron in 3 of 4 ores as high grade products analyzing 74 to 85 percent iron 2 5 to 6 percent manganese and 3 to 10 percent SiO₂ The nonmagnetic fractions were leached with dilute sulfuric acid extracting 83 to 93 percent of the contained manganese and small amounts of iron The leach solutions contained Mn to Fe and Mn to SiO₂ ratios favorable enough for possible consumption in metallurgical manganese operations Exceptionally pure Mn solutions were achieved by pressure precipitation of the remaining iron in an autoclave followed by lime precipitation of silica and alumina

Carolina Science and Math Carolina Biological Supply Company, 2003

Improving Student Comprehension in Chemistry Laboratories Tracy Lynn Haroff, 2006

Chemistry Education Javier García-Martínez, Elena Serrano-Torregrosa, 2015-05-04 Winner of the CHOICE Outstanding Academic Title 2017 Award This comprehensive collection of top level contributions provides a thorough review of the vibrant field of chemistry education Highly experienced chemistry professors and education experts cover the latest developments in chemistry learning and teaching as well as the pivotal role of chemistry for shaping a more sustainable future Adopting a practice oriented approach the current challenges and opportunities posed by chemistry education are critically discussed highlighting the pitfalls that can occur in teaching chemistry and how to circumvent them The main topics discussed include best practices project based education blended learning and the role of technology including e learning and science visualization Hands on recommendations on how to optimally implement innovative strategies of teaching chemistry at university and high school levels make this book an essential resource for anybody interested in either teaching or learning chemistry more effectively from experience chemistry professors to secondary school teachers from educators with no formal training in didactics to frustrated chemistry students

Chemistry Eugene LeMay, Jr., Herbert Beall, Karen M. Robblee, Douglas C. Brower, Prentice-Hall Staff, 2002-02

, Book of Abstracts American Chemical Society. Meeting, American Chemical Society, 1974

Chemistry Frank Jenkins, 1992

Report summaries United States. Environmental Protection Agency, 1983

Elements of Experimental Organic Chemistry Michael McCormick, 2002

Fundamentals of Environmental Sampling and Analysis Chunlong Zhang, 2007-03-09 An integrated approach to understanding the principles of sampling chemical analysis and instrumentation This unique reference focuses on the overall framework and why various methodologies are used in environmental sampling and analysis An understanding of the underlying theories and principles empowers environmental professionals to select and adapt the proper sampling and analytical protocols for specific contaminants as well as for specific project applications Covering both field sampling and laboratory analysis Fundamentals

of Environmental Sampling and Analysis includes A review of the basic analytical and organic chemistry statistics hydrogeology and environmental regulations relevant to sampling and analysis An overview of the fundamentals of environmental sampling design sampling techniques and quality assurance quality control QA QC essential to acquire quality environmental data A detailed discussion of the theories of absorption spectroscopy for qualitative and quantitative environmental analysis metal analysis using various atomic absorption and emission spectrometric methods and the instrumental principles of common chromatographic and electrochemical methods An introduction to advanced analytical techniques including various hyphenated mass spectrometries and nuclear magnetic resonance spectroscopy With real life case studies that illustrate the principles plus problems and questions at the end of each chapter to solidify understanding this is a practical hands on reference for practitioners and a great textbook for upper level undergraduates and graduate students in environmental science and engineering

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