

# STOICHIOMETRY: MASS – MOLE WORKSHEET

Solve the following stoichiometry problems and write the correct answer with units (For example: 34 mol ó 4.62 g) Round up to 2 decimals.

1. Based on the following chemical equation, please answer the following questions:



- a. Calculate the amount of moles of NaCl needed to produce 75g of FeCl<sub>3</sub>.

Answer:

- b. Calculate the amount of Fe<sub>2</sub>(SO<sub>4</sub>)<sub>3</sub> in grams to produce 215 g of Na<sub>2</sub>SO<sub>4</sub>.

Answer:

- c. Calculate the amount in grams of FeCl<sub>3</sub> produced with 95g of Fe<sub>2</sub>(SO<sub>4</sub>)<sub>3</sub>

Answer:

- d. Calculate the amount of moles of Fe<sub>2</sub>(SO<sub>4</sub>)<sub>3</sub> needed to react with 123.5 g of NaCl

Answer:

2. How many moles of CaCO<sub>3</sub> can be produced if we make 12.3g of Ca(HCO<sub>3</sub>)<sub>2</sub> reacts with Ca(OH)<sub>2</sub>?



Answer:

# Stoichiometry Mass To Mass Stoichiometry Problems Answers

**Morris Hein, Susan Arena, Cary Willard**



## **Stoichiometry Mass To Mass Stoichiometry Problems Answers:**

*Microbiology (Questions and Answers), 5e* Purshotam Kaushik & Kirti Kaushik, 2022 Microbiology is an engaging textbook presenting balanced and comprehensive account of major areas of microbiology in the form of questions and answers This question answer approach to present complex topics and theories of microbiology regarding cellular and non cellular microorganisms microbial genetics and molecular biology in higher plants and animals makes the subject interesting and easily comprehensible for the students

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