

Chapter 3

The Mole and Stoichiometry

Multiple Choice Questions

Section 3.1

Difficulty Level: easy

1. An atom of copper has a mass about four times greater than that of an atom of oxygen. What choice makes the correct comparison of the relative numbers of copper and oxygen atoms in 1,000 g of each element?
- The number of copper and oxygen atoms is the same.
 - There are one thousand times as many copper atoms as oxygen atoms.
 - There are one thousand times as many oxygen atoms as copper atoms.
 - There are four times as many oxygen atoms as copper atoms.
 - There are four times as many copper atoms as oxygen atoms.

Answer: d

Section 3.1

Difficulty Level: medium

2. An atom of sulfur is about four times smaller in mass than an atom of iodine. How many grams of neon will contain the same number of atoms as 1,000 g of iodine?
- 4 g S
 - 250 g S
 - 400 g S
 - 1,000 g S
 - 4,000 g S

Answer: b

Section 3.1

Difficulty Level: easy

3. Which of these quantities does not represent 1.00 mol of the indicated substance?
- 6.02×10^{23} C atoms
 - 26.0 g Fe
 - 12.01 g C
 - 65.4 g Zn
 - 6.02×10^{23} Fe atoms

Answer: b

Stoichiometry Multiple Choice Questions

Peterson's



Stoichiometry Multiple Choice Questions:

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