

Stoichiometry Worksheet and Key



1. How many moles of O_2 will be formed from 1.65 moles of KClO_3 ?

| | | |
|--------------------------|---------------------------|--------------------------|
| 1.65 mol KClO_3 | _____ mol O_2 | = _____ mol O_2 |
| | _____ mol KClO_3 | |

2. How many moles of KClO_3 are needed to make 3.50 moles of KCl ?

| | | |
|-----------------------|--|-----------------------------|
| 3.50 mol KCl | | = _____ mol KClO_3 |
| | | |

3. How many moles of KCl will be formed from 2.73 moles of KClO_3 ?

| | | |
|-------|--|---------|
| _____ | | = _____ |
| | | |



4. How many moles of Fe_2O_3 are produced when 0.275 moles of Fe is reacted?

| | | |
|-----------------------|--|-------------------------------------|
| 0.275 mol Fe | | = _____ mol Fe_2O_3 |
| | | |

5. How many moles of Fe_2O_3 are produced when 31.0 moles of O_2 is reacted?

| | | |
|-------|--|---------|
| _____ | | = _____ |
| | | |

6. How many moles of O_2 are needed to react with 8.9 moles of Fe ?

| | | |
|-------|--|---------|
| _____ | | = _____ |
| | | |



7. How many moles of O_2 are produced when 1.26 moles of H_2O is reacted?

8. How many moles of H_2O are needed to produce 55.7 moles of H_2 ?

9. If enough H_2O is reacted to produce 3.40 moles of H_2 , then how many moles of O_2 must have been made?
(a bit challenging, but just think about it and you can probably figure it out)

Stoichiometry Problems Worksheet Answer Key

**Frank Schaffer Publications, Joan
DiStasio**



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