

This print-out should have 16 questions. Multiple-choice questions may continue on the next column or page – find all choices before answering.

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**001 10.0 points**

Which compound has the wrong chemical formula?

1.  $\text{Mg}(\text{OH})_2$
2.  $\text{Ba}_3(\text{PO}_4)_2$
3.  $\text{CaOH}$  correct
4.  $(\text{NH}_4)_2\text{SO}_4$

**Explanation:**

The calcium ion is  $\text{Ca}^{2+}$ ; the hydroxide ion is  $\text{OH}^-$ . Two  $\text{OH}^-$  are needed to balance the charge of each  $\text{Ca}^{2+}$ , so the formula is  $\text{Ca}(\text{OH})_2$ .

The magnesium ion is  $\text{Mg}^{2+}$ ; the hydroxide ion is  $\text{OH}^-$ . Two  $\text{OH}^-$  are needed to balance the charge of each  $\text{Mg}^{2+}$ , so the formula is  $\text{Mg}(\text{OH})_2$ .

The barium ion is  $\text{Ba}^{2+}$ ; the phosphate ion is  $\text{PO}_4^{3-}$ . Two  $\text{PO}_4^{3-}$  are needed to balance the charge of every three  $\text{Ba}^{2+}$ . (This gives a total anion charge of  $-6$  and a total cation charge of  $+6$ .) The formula is  $\text{Ba}_3(\text{PO}_4)_2$ .

The ammonium ion is  $\text{NH}_4^+$ ; the sulfate ion is  $\text{SO}_4^{2-}$ . Two  $\text{NH}_4^+$  are needed to balance the charge of each  $\text{SO}_4^{2-}$ , so the formula is  $(\text{NH}_4)_2\text{SO}_4$ .

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**002 10.0 points**

Which one has the greatest number of atoms?

1. 3.05 moles of water
2. 3.05 moles of helium
3. All have the same number of atoms
4. 3.05 moles of  $\text{CH}_4$  correct
5. 3.05 moles of argon

**Explanation:**

For 3.05 moles of water:

$$\begin{aligned} ? \text{ atoms} &= 3.05 \text{ mol H}_2\text{O} \times \frac{6.02 \times 10^{23} \text{ molec}}{1 \text{ mol}} \\ &\quad \times \frac{3 \text{ atoms}}{1 \text{ molecule}} \\ &= 5.51 \times 10^{24} \text{ atoms} \end{aligned}$$

For 3.05 moles of  $\text{CH}_4$ :

$$\begin{aligned} ? \text{ atoms} &= 3.05 \text{ mol CH}_4 \times \frac{6.02 \times 10^{23} \text{ molec}}{1 \text{ mol}} \\ &\quad \times \frac{5 \text{ atoms}}{1 \text{ molecule}} \\ &= 9.18 \times 10^{24} \text{ atoms} \end{aligned}$$

For 3.05 moles of helium:

$$\begin{aligned} ? \text{ atoms} &= 3.05 \text{ mol He} \times \frac{6.02 \times 10^{23} \text{ atoms}}{1 \text{ mol}} \\ &= 1.84 \times 10^{24} \text{ atoms} \end{aligned}$$

For 3.5 moles of argon:

$$\begin{aligned} ? \text{ atoms} &= 3.05 \text{ mol Ar} \times \frac{6.02 \times 10^{23} \text{ atoms}}{1 \text{ mol}} \\ &= 1.84 \times 10^{24} \text{ atoms} \end{aligned}$$

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**003 10.0 points**

If 100.0 grams of copper (Cu) completely reacts with 25.0 grams of oxygen, how much copper(II) oxide ( $\text{CuO}$ ) will form from 140.0 grams of copper and excess oxygen? (Note:  $\text{CuO}$  is the only product of this reaction.)

1. 35.0 g
2. 160.0 g
3. 150.0 g
4. 175.0 g correct
5. 200.0 g

**Explanation:**

$$\begin{aligned} m_{\text{Cu, ini}} &= 100.0 \text{ g} & m_{\text{O}_2} &= 25.0 \text{ g} \\ m_{\text{Cu, fin}} &= 140.0 \text{ g} \end{aligned}$$

If 100 g copper and 25 g oxygen react completely with each other, there must be 125 g of product formed (law of conservation of mass). This product is  $\text{CuO}$ .

# Stoichiometry Quest Answers

**TD Snyder**



## **Stoichiometry Quest Answers:**

**The Quest for Food** Harald Brüssow, 2007-05-11 This book explores the links between food and human cultural and physical evolution Each chapter begins by summarizing the basic knowledge in the field discusses recent research results and confirms or challenges established concepts inviting new insight and provoking new questions This book catalyzes discussion between scientists working on one side in food science and on the other side in biological and biomedical research

**Stoichiometry and Materials Science** Alessio Innocenti, Norlida Kamarulzaman, 2012-04-11 The aim of this book is to provide an overview on the importance of stoichiometry in the materials science field It presents a collection of selected research articles and reviews providing up to date information related to stoichiometry at various levels Being materials science an interdisciplinary area the book has been divided in multiple sections each for a specific field of applications The first two sections introduce the role of stoichiometry in nanotechnology and defect chemistry providing examples of state of the art technologies Section three and four are focused on intermetallic compounds and metal oxides Section five describes the importance of stoichiometry in electrochemical applications In section six new strategies for solid phase synthesis are reported while a cross sectional approach to the influence of stoichiometry in energy production is the topic of the last section Though specifically addressed to readers with a background in physical science I believe this book will be of interest to researchers working in materials science engineering and technology

**Complex Ecology** Charles G. Curtin, Timothy F. H. Allen, 2018-05-31 Research papers from the end of twentieth century have been assembled alongside expert commentary for the first collected volume on complexity based ecology

**Questions and Answers in Environmental Science** S.K. Basu, Amit Krishna De, 2005 The Sustainable Future Of Humany Lies In Understanding The Earth And Its Environment For This Reason Environmental Science Has A Purview That Overlaps Several Other Disciplines From Biology To Economics Geology To Sociology Every Subject Has A Significant Relationship With Some Area Of Environmental Science However It Is Often Difficult Time Consuming And Exhaustive To Keep Pace With New Trends In Such A Broad Based Field

**Foundations of College Chemistry, Laboratory** Morris Hein, Susan Arena, 2010-08-09 Learning the fundamentals of chemistry can be a difficult task to undertake for health professionals For over 35 years this book has helped them master the chemistry skills they need to succeed It provides them with clear and logical explanations of chemical concepts and problem solving They ll learn how to apply concepts with the help of worked out examples In addition Chemistry in Action features and conceptual questions checks brings together the understanding of chemistry and relates chemistry to things health professionals experience on a regular basis

**A Cultural History of Chemistry in the Nineteenth Century** Peter J. Ramberg, 2023-12-14 A Cultural History of Chemistry in the Nineteenth Century covers the period from 1815 to 1914 and the birth of modern chemistry The elaboration of atomic theory and new ideas of periodicity structure bonding and equilibrium emerged in tandem with new instruments and practices The chemical industry expanded exponentially fuelled by

an increasing demand for steel aluminium dyestuffs pharmaceuticals and consumer goods And the chemical laboratory became established in its two distinct modern settings of the university and industry At the turn of the century the discovery of radioactivity took hold of the public imagination drawing chemistry closer to physics even as it threatened to undermine the whole concept of atomism The 6 volume set of the Cultural History of Chemistry presents the first comprehensive history from the Bronze Age to today covering all forms and aspects of chemistry and its ever changing social context The themes covered in each volume are theory and concepts practice and experiment laboratories and technology culture and science society and environment trade and industry learning and institutions art and representation Peter J Ramberg is Professor of the History of Science at Truman State University USA Volume 5 in the Cultural History of Chemistry set General Editors Peter J T Morris University College London UK and Alan Rocke Case Western Reserve University USA

**Ionic Liquids in Synthesis** Peter Wasserscheid, Thomas Welton, 2006-03-06 The demand for increasingly clean and efficient chemical syntheses is becoming more urgent from both an economic and an environmental standpoint Many technologies rely on large quantities of hazardous even toxic solvents A promising and now established approach is the development of new ionic solvents that are fluid at room temperature These solvents not only have the potential to increase chemical reactivity and thus lead to more efficient processes but are also non flammable and are less toxic than conventional solvents due to their low vapor pressure This volume brings together the latest developments in this fascinating field supplemented by numerous practical tips and thus provides those working in both research and industry with an indispensable source of information

*The Encyclopedia of Biochemistry* Roger John Williams, Edwin M. Lansford, 1967 For the non specialist Articles on broader topics presuppose only general scientific background specialized topics are more technical The alphabetically arranged articles cover general and specific topics in chemistry physics methodology metabolism nutrition diseases and disorders as related to biochemistry as well as short biographies and more peripheral topics Longer articles have appended references Indexed *Molecular Toxicology* P. David Josephy, Bengt Mannervik, 2006 The science of toxicology has progressed considerably since *Molecular Toxicology* was first published in 1997 New advances in biochemical and molecular biological experimental techniques have helped researchers understand the precise effects of toxins and foreign compounds on living things at the molecular cellular and organismal levels Breakthrough research has recently been completed illuminating the human genome and the role of enzymes in toxic biochemical reaction mechanisms Toxicology now covers drug metabolism and design carcinogenesis programmed cell death and DNA repair among other subjects The second edition captures these and other advances and broadens its scope to address the experimental science of toxicology The first edition of *Molecular Toxicology* has become an indispensable resource for graduate students in molecular and biochemical toxicology courses as well as academic researchers and industrial researchers in toxicology Rigorously updated and revised the new edition commands an unrivaled authority in the field of molecular toxicology **Proceedings**, 1968 Ionic

Liquids in Synthesis, 2 Volume Set Peter Wasserscheid, Thomas Welton, 2008 The second completely revised and enlarged edition of what has become the standard reference work in this fascinating field brings together the latest developments supplemented by numerous practical tips providing those working in both research and industry with an indispensable source of information New contributions have been added to reflect the fact that industrial processes are already established and ionic liquids are now commercially available A must for everyone working in the field **Proceedings of the ...**

**Southern Water Resources and Pollution Control Conference**, 1965 The Journal of Experimental Biology, 2009-06

*Multi-Scale Biogeochemical Processes in Soil Ecosystems* Yu Yang, Marco Keiluweit, Nicola Senesi, Baoshan Xing, 2022-04-12 MULTI SCALE BIOGEOCHEMICAL PROCESSES IN SOIL ECOSYSTEMS Provides a state of the art overview of research in soil biogeochemical processes and strategies for greenhouse gas mitigation under climate change Food security and soil health for the rapidly growing human population are threatened by increased temperature and drought soil erosion and soil quality degradation and other problems caused by human activities and a changing climate Because greenhouse gas emission is the primary driver of climate change a complete understanding of the cycles of carbon and major nutritional elements is critical for developing innovative strategies to sustain agricultural development and environmental conservation Multi Scale Biogeochemical Processes in Soil Ecosystems Critical Reactions and Resilience to Climate Changes is an up to date overview of recent research in soil biogeochemical processes and applications in ecosystem management Organized into three parts the text examines molecular scale processes and critical reactions presents ecosystem scale studies of ecological hotspots and discusses large scale modeling and prediction of global biogeochemical cycles Part of the Wiley IUPAC Series on Biophysico Chemical Processes in Environmental Systems this authoritative volume Provides readers with a systematic and interdisciplinary approach to sustainable agricultural development and management of soil ecosystems in a changing climate Features contributions from an international team of leading scientists Examines topics such as soil organic matter stabilization soil biogeochemistry modeling and soil responses to environmental changes Discusses strategies for mitigating greenhouse gas emission and improving soil health and ecosystems resilience Includes an introduction to working across scales to project soil biogeochemical responses to climatic change Multi Scale Biogeochemical Processes in Soil Ecosystems Critical Reactions and Resilience to Climate Changes is essential reading for scientists engineers agronomists chemists biologists academic researchers consultants and other professionals whose work involves the nutrient cycle ecosystem management and climate change **The Software Encyclopedia**, 1988 **Extracellular Vesicles** Wojciech Chrzanowski, Chwee Teck Lim, Sally Yunsun Kim, 2021-10-20 Extracellular and biofluids vesicles EVs are highly specialised yet ubiquitous nanoscale messengers secreted by cells With the development of stem cell engineering EVs promise to deliver next generation tools in regenerative medicine and tissue engineering as well as in diagnostics A vibrant and promising field this book provides the first resource to the field Covering basic cell biology including EV production and

intracellular communication this book will provide material scientists and engineers with a foundation to the necessary biology The reader will then learn about the isolation of extracellular vesicles their physicochemical characterisation and therapeutic application of EVs in regenerative medicine as well as their potential as biomarkers in medical diagnostic This book will also discuss the regulatory landscape of EVs Bridging cell biology biomaterials biophysics and biomedical engineering the content of this book is written with a broad interdisciplinary audience in mind Researchers new and established will find this a must have on their shelf

**Green Chemical Processes** Mark Anthony Benvenuto, 2017-10-10  
The greening of industry processes i.e making them more sustainable is a popular and often lucrative trend which has seen increased attention in recent years Green Chemical Processes the 2nd volume of Green Chemical Processing covers the hot topic of sustainability in chemistry with a view to education as well as considering corporate and environmental interests e.g in the context of energy production The diverse team of authors allows for a balance between these different but interconnected perspectives The American Chemical Society's 12 Principles of Green Chemistry are woven throughout this text as well as the series to which this book belongs

The Publishers' Trade List Annual, 1974      **Technology in the Curriculum: Science resource guide**, 1986      **Iterations** John W. Moore, 1981

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