WRITING FORMULAS FROM NAMES

Name <u>Key</u>

Write the formulas of the following compounds.

- 1. ammonium phosphate (NH4) 3 PO4
- 3. Iron (III) oxide Fe_2O_3
- 4. carbon monoxide ______
- 5. calcium chloride Ca C/2
- 7. magnesium hydroxide $M_g(OH)_2$
- 8. aluminum sulfate $Al_2(50_4)_3$
- 9. copper (II) sulfate Cu 504
- 10. lead (IV) chromate $Pb_{2}(CrO_{4})_{4} \Rightarrow Pb(CrO_{4})_{2}$
- 12. potassium permanganate KMnOy
- sodium hydrogen carbonate

 Na HCO₃
- 14. zinc nitrate $Z_{3}(NO_{3})_{2}$
- 15. aluminum sulfite $A/_{2}(5O_{3})_{3}$

David C. McMillin

Science Spectrum Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2003-03 Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science, 2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction Carolina Science and Math Carolina Biological Supply Company, 2003 Chemistry Math Concepts David C. McMillin, 2001 Chemistry McGraw-Hill Staff, 2001-05-01

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