# Section 1 Review page 118

1. Describe Democritus's atomic theory.

Everything in the universe is made up of atoms.

- 2. Summarize the main ideas of Dalton's theory.
  - Elements are made up of tiny unique particles called atoms.
  - b. Atoms cannot be divided.
  - c. Atoms of the same element are identical.
  - d. Atoms of different elements can join to form molecules.
- Explain why Dalton's theory was more successful tan Democritus's theory.
   Dalton had experimental evidence to support his theory, while Democritus did not.
- Compare Thomson's atomic model with Rutherford's atomic model.
   Thomson's model- electrons were embedded in a sphere of positive charge Rutherford's model- positive charge is concentrated in the nucleus and electrons surrounded the nucleus.
- 5. Analyzing elements
  - a. How did the cathode-ray tube experiment lead to the conclusion that atoms contain electrons? The air was removed from the tube and the beam of negative particles was observed. This suggests that atoms are made up of smaller units with negative charges.
  - b. How did the gold-foil experiment lead to the conclusion that the atom has a nucleus? the positive particles were deflected at large angles and therefore this suggests that a positive charge is concentrated in the center of the atom. (nucleus)
- Making Inferences: Does the term indivisible still describe the atom? Explain. No, the atom is not indivisible. It is made up of smaller parts such as the electron.

# Section 2 review page 127

- List the charge, mass, and location of each of the three subatomic particles found in atoms. The proton is positive, the neutron is neutral and electrons are negative. Protons and neutrons are found in the nucleus of an atom. The mass of a proton 1850 times greater than that of an electron.
- 2. Explain how you can use an atom's mass number and atomic number to determine the number of protons, electrons, and neutrons to determine the number of protons, electrons, and neutrons in the atom. The atomic number equals the number of protons. The number of electrons equals the number of protons. To find the number of neutrons subtract the atomic number from the mass number.
- Identify the subatomic particle used to define an element, and explain why
  this particle is used. The proton defines the element's number because the each
  element always has the same number of protons, while the number of neutrons
  and electrons can vary.
- Explain why the masses of atoms of the same element may differ. Because they can have different number of neutrons. For example Hydrogen (H) H, H<sub>1</sub>Deterium, H<sub>2</sub> Tritium. These are called isotopes.

Michael B. Leyden

ENC Focus .2000 Communicating Science: A Practical Guide For Engineers And Physical Scientists Edith S Boxman, Reuven (Raymond) L Boxman, 2016-12-28 The book under review is indeed a guide and a practical one too but it is not only for engineers and physical scientists it is useful to all those who communicate technical information or need to and want to do the job well The book achieves the objectives every PhD student may not want to read the book but should and will not regret it European Science EditingRead this book before you write your thesis or journal paper Communicating Science is a textbook and reference on scientific writing oriented primarily at researchers in the physical sciences and engineering It is written from the perspective of an experienced researcher It draws on the authors experience of teaching and working with both native English speakers and English as a Second Language ESL writers For the range of topics covered this book is relatively short and tersely written in order to appeal to busy researchers Graduate students and early career researchers will be guided through the researcher's basic communication tasks writing theses journal papers and internal reports presenting lectures and posters and preparing research proposals Extensive best practice examples and analyses of common problems are presented Advanced researchers who aim to commercialize their research results will be introduced to business plans and patents so that they can communicate optimally with patent attorneys and business analysts Likewise advanced researchers will be assisted in conveying the results of their research to the industrial and business community governmental circles and the general public in the chapter on popular media Researchers at all levels will find the chapter on CV s and job hunting helpful The Writing Well chapter will assist researchers to improve their English usage in scientific writing This chapter is oriented both at native English speakers who have an intuitive command of English but often lack formal instruction on grammar and structure and non native English writers who often have had formal instruction but lack intuitive grasp of what sounds good Mentors will find the book a useful tool for systematically guiding their students in their early writing efforts If your students read this book first you will save time Communicating Science may serve as a textbook for graduate level courses in scientific writing Science Spectrum Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2003-03 ASVAB 2017-2018 Strategies, Practice & Review with 4 Practice Tests Kaplan Test Prep, 2016-07-05 Offers test taking tips diagnostic tests review and three practice exams with answers and detailed explanations for all portions of the ASVAB Resources for Teaching Middle School Science Smithsonian Institution, National Academy of Engineering, National Science Resources Center of the National Academy of Sciences, Institute of Medicine, 1998-04-30 With age appropriate inquiry centered curriculum materials and sound teaching practices middle school science can capture the interest and energy of adolescent students and expand their understanding of the world around them Resources for Teaching Middle School Science developed by the National Science Resources Center NSRC is a valuable tool for identifying and selecting effective science curriculum materials that will engage students

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others have altered their emphasis along with their titles some have been combined under a new name whereas others have had to be discontinued The current list of Specialist Periodical Reports can be seen on the inside flap of this volume Saturday Review of Politics, Literature, Science and Art, 1865 Addison-Wesley Introduction to Physical Science Michael B. Levden, 1988 Bulletin of the Atomic Scientists, 1972-03 The Bulletin of the Atomic Scientists is the premier public resource on scientific and technological developments that impact global security Founded by Manhattan Project Scientists the Bulletin's iconic Doomsday Clock stimulates solutions for a safer world Athenaeum ,1865 The Saturday Review of Politics, Literature, Science, Art, and Finance, 1865 Marine Eutrophication in Perspective Folkert de Jong, 2007-04-21 Is ecological knowledge relevant for environmental policy and if so to what extent and in what way These are some of the basic guestions addressed in this book Triggered by a series of oxygen depletion events in German Danish and Swedish coastal waters at the beginning of the 1980s international policies to reduce inputs of nitrogen and phosphorus to the North Sea were agreed upon by the North Sea states The book provides a critical analysis of the role scientists and scientific information as well as civil servants have played in the formulation and implementation of these decisions Education During and Beyond COVID-19 C. Raj Kumar, Mousumi Mukherjee, Tatiana Belousova, Nisha Nair, 2022-04-27 This book offers insights into how higher educational institutions and educators have responded to the immense challenges of managing the COVID 19 pandemic Written by global experts in the field of higher education it offers a multidimensional overview of the digital transformation governance and social justice issues within higher education institutions during the pandemic It provides theoretical insights and conceptual analysis of the emerging trends in global higher education the challenges and possible ways to address them to shape more sustainable qualitative and socially equitable higher education for future generations The book appeals to academics and students engaged in the education community **Mathematical Reviews** ,2005 Saturday Review of Politics, Literature, Science and Art ,1865

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