

CHEMISTRY FINAL EXAM REVIEW GUIDE 2012 Scott

DISCLAIMER: THIS IS A SAMPLE OF QUESTIONS/PROBLEMS FROM THIS YEAR. PLEASE BE SURE TO REVIEW YOUR NOTES, LABS, COJ'S, OBJECTIVE SHEETS, ETC. DO NOT RELY ONLY ON THIS REVIEW GUIDE.

Multiple Choice

Identify the choice that best completes the statement or answers the question.

- _____ 1) Which of the following observations is quantitative?
a. The liquid turns blue litmus paper red.
b. The liquid boils at 100°C.
c. The liquid tastes bitter.
d. The liquid is cloudy.
- _____ 2) The symbols for units of length in order from largest to smallest are
a. m, cm, mm, km. c. km, mm, cm, m.
b. mm, m, cm, km. d. km, m, cm, mm.
- _____ 3) A volume of 1 milliliter is equivalent to
a. 1 cubic centimeter. c. 1 liter.
b. 1 gram. d. 10^{-3} cubic decimeters.
- _____ 4) Which pair of quantities determines the density of a material?
a. mass and weight c. volume and concentration
b. volume and weight d. volume and mass
- _____ 5) A sample of gold has a mass of 96.5 g and a volume of 5.00 cm³. The density of gold is
a. 0.0518 g/cm³. c. 101.5 g/cm³.
b. 19.3 g/cm³. d. 483 g/cm³.
- _____ 6) The number of grams equal to 0.5 kg is
a. 0.0005. c. 500.
b. 0.005. d. 5000.
- _____ 7) If 1 inch equals 2.54 cm, how many centimeters equal 1 yard?
a. 0.0706 cm c. 30.5 cm
b. 14.2 cm d. 91.4 cm
- _____ 8) A chemist obtained the values 5.2246 g, 5.2353 g, and 5.2501 g for the mass of a sample. Without knowing the true mass of the sample, it can be said that these values have
a. good precision.
b. good accuracy.
c. poor precision.
d. poor accuracy.
- _____ 9) The measurement 0.035550 g rounded off to two significant figures would be
a. 0.03 g. c. 0.036 g.
b. 0.35 g. d. 3.5×10^2 g.
- _____ 10) When 64.4 is divided by 2.00, the correct number of significant figures in the result is

Scope For Chemistry Final Examination

**Nellie Mary Apanasewicz, Seymour
Michael Rosen, United States. Office of
Education**

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Federal Register, 2013-04 **Cumulative Supplement to the Code of Federal Regulations of the United States of America**, 1943 The Code of Federal Regulations of the United States of America Having General Applicability and Legal Effect in Force June 1, 1938, 1939 **General Orders** United States. War Department, 1910 *Supplement to the Code of Federal Regulations of the United States of America*, 1939 **University of Michigan Official Publication** University of Michigan, 1973 Each number is the catalogue of a specific school or college of the University *Special Regulations* United States. War Department, 1917 **Traceability in Chemical Measurement** Paul De Bièvre, Helmut Günzler, 2005-12-06

Metrological traceability of chemical measurement results means the establishment of a relation to metrological stated references through an unbroken chain of comparisons This volume collects 56 outstanding papers on the topic mostly published in the period 2000-2003 in the journal Accreditation and Quality Assurance They provide the latest understanding and possibly the rationale why it is important to integrate the concept of metrological traceability including suitable measurement standards such as certified reference materials into the standard measurement procedures of every analytical laboratory In addition this anthology considers the benefits to both the analytical laboratory and the user of the measurement results *Final Examinations in the Russian Ten-year School* Nellie Mary Apanasewicz, Seymour Michael Rosen, United States. Office of Education, 1966 **Bulletin**, 1959 Statistics of Land-grant Colleges and Universities United States. Office of Education, 1954 Teaching Rapid and Slow Learners in High Schools Romaine Prior Mackie (1898-), Abul Hassan K. Sassani, Clayton D. Hutchins, Elizabeth (Nelson) Layton, Gertrude Minnie Lewis, Henry Horton Armsby, Nora Ernestine Beust, United States. Office of Education, Wilhelmina Hill, Helen Katherine Mackintosh, Lloyd M. Dunn, 1954 *Bulletin* United States. Office of Education, 1954 Graduate School Announcements for ... University of North Dakota. Graduate School, 1961 *Chemistry of Cement*, 1962 **Bulletin - Bureau of Chemistry** United States. Bureau of Chemistry, 1887

Index of General Orders and Circulars United States. War Dept, 1912 Tietz Textbook of Clinical Chemistry and Molecular Diagnostics - E-Book Nader Rifai, 2017-01-16 The Tietz Textbook of Clinical Chemistry and Molecular Diagnostics 6th Edition provides the most current and authoritative guidance on selecting performing and evaluating the results of new and established laboratory tests This classic clinical chemistry reference offers encyclopedic coverage detailing everything you need to know including analytical criteria for the medical usefulness of laboratory tests variables that affect tests and results laboratory medicine applications of statistical methods and most importantly clinical utility and interpretation of laboratory tests It is THE definitive reference in clinical chemistry and molecular diagnostics now fully searchable and with quarterly content updates podcasts clinical cases animations and extended content online through Expert Consult Analytical criteria focus on the medical usefulness of laboratory procedures Reference ranges show new approaches for establishing these ranges and provide the latest information on this topic Lab management and costs gives students and chemists the

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