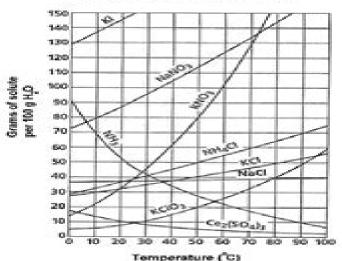
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- Using the solubility curves above, answer the following questions:
 - a. Which substance will form a saturated solution when 35 g is dissolved in 100 g of water at 75°C? Answer in a complete sentence.

tassium chilorate (KClOs) will form a b. Which substance(s) on the graph are gases? Justify your response using complete sentences.

NH3 is a gas. The solubility of gases dichease as the temprature increases. Since the slope of the curve for NH3 is regative, this indicates a decrease in solubility with increased

c. Which substance is LEAST affected by temperature? Justify your response using complete

odiam chloride (NaCl) is least affected by temperature. This is indicated in the small slupe of its culive.

d. If 65 g of ammonium chloride is dissolved in 100 g of water at 90°C, will the solution be saturated, unsaturated or supersaturated. Justify your response using complete sentences.

The solution will be unsaturated the point 659 of NH4Clat 90°C falls below the curve indicating that mare solute could be dissolved

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Ben Piper, David Clinton

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