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545/2
CHEMISTRY
Paper 2
Oct./Nov. 2018
2 hours.



UGANDA NATIONAL EXAMINATIONS BOARD

Uganda Certificate of Education

CHEMISTRY

Paper 2

2 hours

INSTRUCTIONS TO CANDIDATES:

Section A consists of 10 structured questions. Answer all questions in this section. Answers to these questions must be written in the spaces provided.

Section B consists of 4 semi-structured questions. Answer any two questions from this section. Answers to the questions must be written in the answer booklet(s) provided.

In both sections all working must be clearly shown.

Where necessary use;

$[H = 1; C = 12; N = 14; O = 16; Na = 23; S = 32; Cl = 35.5]$

1 mole of gas occupies 24 l at room temperature.

1 mole of gas occupies 22.4 l at s.t.p.

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