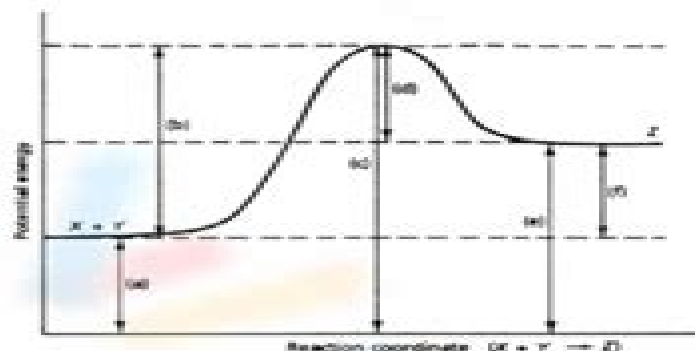
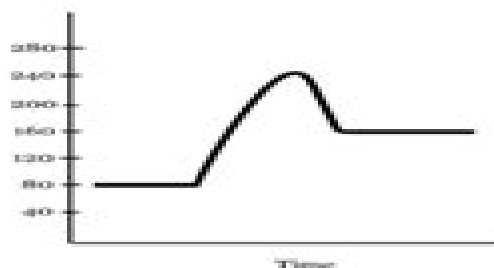


Potential Energy Diagram Worksheet **ANSWERS**



1. Which of the letters a-f in the diagram represents the potential energy of the products? **e**
2. Which letter indicates the potential energy of the activated complex? **c**
3. Which letter indicates the potential energy of the reactants? **a**
4. Which letter indicates the activation energy? **b**
5. Which letter indicates the heat of reaction? **f**
6. Is the reaction exothermic or endothermic? **endo**

7. Which letter indicates the activation energy of the reverse reaction? **d**
8. Which letter indicates the heat of reaction of the reverse reaction? **f**
9. Is the reverse reaction exothermic or endothermic? **exo**



(endothermic or exothermic).

1. The PE of the reactants of the forward reaction is about **80** kilojoules.
2. The PE of the products of the forward reaction is about **160** kilojoules.
3. The PE of the activated complex of the forward reaction is about **240** kilojoules.
4. The activation energy of the forward reaction is about **160** kilojoules.
5. The heat of reaction (ΔH) of the forward reaction is about **+80** kilojoules.
6. The forward reaction is **endothermic**
7. The PE of the reactants of the reverse reaction is about **160** kilojoules.
8. The PE of the products of the reverse reaction is about **80** kilojoules.
9. The PE of the activated complex of the reverse reaction is about **240** kilojoules.
10. The activation energy of the reverse reaction is about **80** kilojoules.
11. The heat of reaction (ΔH) of the reverse reaction is about **-80** kilojoules.
12. The reverse reaction is **exothermic** (endothermic or exothermic).

Potential Energy Diagram Worksheet

L Manion

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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Chemistry Carson-Dellosa Publishing, 2015-03-16 Chemistry for grades 9 to 12 is designed to aid in the review and practice of chemistry topics Chemistry covers topics such as metrics and measurements matter atomic structure bonds compounds chemical equations molarity and acids and bases The book includes realistic diagrams and engaging activities to support practice in all areas of chemistry The 100 Series science books span grades 5 to 12 The activities in each book reinforce essential science skill practice in the areas of life science physical science and earth science The books include engaging grade appropriate activities and clear thumbnail answer keys Each book has 128 pages and 100 pages or more of reproducible content to help students review and reinforce essential skills in individual science topics The series will be aligned to current science standards

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Waldeck,Horst-Dieter Försterling,2024-10-25 This is a Solutions Manual to Accompany with solutions to the exercises in the main volume of Principles of Physical Chemistry Third Edition This book provides a unique approach to introduce undergraduate students to the concepts and methods of physical chemistry which are the foundational principles of Chemistry The book introduces the student to the principles underlying the essential sub fields of quantum mechanics atomic and molecular structure atomic and molecular spectroscopy statistical thermodynamics classical thermodynamics solutions and equilibria electrochemistry kinetics and reaction dynamics macromolecules and organized molecular assemblies Importantly the book develops and applies these principles to supramolecular assemblies and supramolecular machines with many examples from biology and nanoscience In this way the book helps the student to see the frontier of modern physical chemistry developments The book begins with a discussion of wave particle duality and proceeds systematically to more complex chemical systems in order to relate the story of physical chemistry in an intellectually coherent manner The topics are organized to correspond with those typically given in each of a two course semester sequence The first 13 chapters present quantum mechanics and spectroscopy to describe and predict the structure of matter atoms molecules and solids Chapters 14 to 29 present statistical thermodynamics and kinetics and applies their principles to understanding equilibria chemical transformations macromolecular properties and supramolecular machines Each chapter of the book begins with a simplified view of a topic and evolves to more rigorous description in order to provide the student and instructor flexibility to choose the level of rigor and detail that suits them best The textbook treats important new directions in physical chemistry research including chapters on macromolecules principles of interfaces and films for organizing matter and supramolecular machines as well as including discussions of modern nanoscience spectroscopy and reaction dynamics throughout the text

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Computational Quantum Chemistry Charles M. Quinn, 2002-02-28 Computational Quantum Chemistry removes much of the mystery of modern computer programs for molecular orbital calculations by showing how to develop Excel spreadsheets to perform model calculations and investigate the properties of basis sets Using the book together with the CD ROM provides a unique interactive learning tool In addition because of the integration of theory with working examples on the CD ROM the reader can apply advanced features available in the spreadsheet to other applications in chemistry physics

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Watkins,2014-03-26 Fundamental Biomechanics of Sport and Exercise is an engaging and comprehensive introductory textbook that explains biomechanical concepts from first principles showing clearly how the science relates to real sport and exercise situations The book is divided into two parts The first provides a clear and detailed introduction to the structure and function of the human musculoskeletal system and its structural adaptations essential for a thorough understanding of human movement The second part focuses on the biomechanics of movement describing the forces that act on the human body and the effects of those forces on the movement of the body Every chapter includes numerous applied examples from sport and exercise helping the student to understand how mechanical concepts describe both simple and complex movements from running and jumping to pole vaulting or kicking a football In addition innovative worksheets for field and laboratory work are included that contain clear objectives a description of method data recording sheets plus a set of exemplary data and worked analysis Alongside these useful features are definitions of key terms plus review questions to aid student learning with detailed solutions provided for all numerical questions No other textbook offers such a clear easy to understand introduction to the fundamentals of biomechanics This is an essential textbook for any biomechanics course taken as part of degree programme in sport and exercise science kinesiology physical therapy sports coaching or athletic training

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