#### Reaction Rates and Potential Energy Diagrams

 Chemical reactions occur when reactants collide. For what reasons may a collision fail to produce a chemical reaction?

Not enough energy; improper angle.

If every collision between reactants leads to a reaction, what determines the rate at which the reaction occurs?

Nature of reactants, Concentration, Temperature, Catalysts.

3. What is the activation energy of a reaction, and how is this energy related to the activated complex of the reaction?

Ea is the minimum amount of energy for a reaction to occur. It is the amount of energy required to create an activated complex.

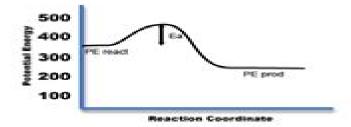
4. What happens when a catalyst is used in a reaction?

A catalyst changes the reaction mechanism, in the process lowering the activation energy.

Name 4 things that will speed up or slow down a chemical reaction.

Increase concentration by distillation of a solvent, Increase concentration by increasing pressure of a gas, Increase temp, Add a catalyst, Add an inhibitor.

 Draw an energy diagram for a reaction. Label the axis, PE of reactants = 350 KJ/mol, Ea = 100 KJ/mol, PE of products = 250 KJ/mol.



Is the reaction in # 6 exothermic or endothermic? Explain.

Exothermic. The AH is -100 KJ/mol which means heat is released.

8. How could you lower the activation energy for the reaction in #6?

Add a catalyst.

## **Potential Energy Diagram Instructioal Fair Answers**

**LL Leslie** 

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In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is really astonishing. Within the pages of "**Potential Energy Diagram Instructioal Fair Answers**," an enthralling opus penned by a highly acclaimed wordsmith, readers set about an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve into the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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