

Solving Thermodynamics Problems

Solving thermodynamic problems can be made significantly easier by using the following procedure:

1. Summarize given data in own words, leave out unneeded information
2. Clearly understand/identify what is being asked for – draw a sketch showing interactions/states and identify a solution strategy. Keep in mind that a multi-step approach may be the easiest
3. Define system boundaries, noting if it is an open or closed system
4. Fix the known states on a P-v/T-v/T-s diagram, using given information
5. Apply conservation of mass to process

- For a control mass/closed system:

$$m_2 - m_1 = 0$$

- For a control volume/open system:

$$\frac{dm}{dt} = \sum \dot{m}_i - \sum \dot{m}_e$$

6. Apply conservation of energy to process (1st law of thermodynamics)

- For a control mass/closed system:

$$\Delta E = Q - W \quad (\text{equilibrium process})$$

$$\frac{dE}{dt} = \dot{Q} - \dot{W} \quad (\text{rate form})$$

- For a control volume/open system:

$$\frac{dE_{cv}}{dt} = \dot{Q}^* - \dot{W}^* + \sum \dot{m}_i \left(h + \frac{V^2}{2} + gz \right)_i - \sum \dot{m}_e \left(h + \frac{V^2}{2} + gz \right)_e$$

7. Solve algebraically for desired quantity using combination of mass balance, energy balance, and definitions (like mass flow rate, volume, etc.)
8. Perform heat transfer analysis to get Q, if necessary (use catalog of heat transfer or steady flow device in appendix)
9. Perform work analysis to get W, if necessary (use catalog of work or steady flow device in appendix)
10. Find properties using Tables or models, such as ideal gas relationship
11. Substitute numbers into equation and solve for desired quantity

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Thermodynamics Problem Solver The Editors of REA, Ralph Pike, 2013-01-01 REA's Thermodynamics Problem Solver Each Problem Solver is an insightful and essential study and solution guide chock full of clear concise problem solving gems Answers to all of your questions can be found in one convenient source from one of the most trusted names in reference solution guides More useful more practical and more informative these study aids are the best review books and textbook companions available They're perfect for undergraduate and graduate studies This highly useful reference provides thorough coverage of pressure work and heat energy entropy first and second laws ideal gas processes vapor refrigeration cycles mixtures and solutions For students in engineering physics and chemistry *Solutions Manual for an Introduction to Thermodynamics* Y.V.C. Rao, 2005-02 This manual contains the complete solution for all the 505 chapter end problems in the textbook *An Introduction to Thermodynamics* and will serve as a handy reference to teachers as well as students The data presented in the form of tables and charts in the main textbook are made use of in this manual for solving the problems

Solved Problems in Thermodynamics Adrian Bejan, Henry Martyn Paynter, 1976 Problems And Solutions On Thermodynamics And Statistical Mechanics (the Volume Is Divided Into Two Parts), 2001 **Thermodynamics Problem Solving in Physical Chemistry** Kathleen E. Murphy, 2020-03-23 Thermodynamics Problem Solving in Physical Chemistry Study Guide and Map is an innovative and unique workbook that guides physical chemistry students through the decision making process to assess a problem situation create appropriate solutions and gain confidence through practice solving physical chemistry problems The workbook includes six major sections with 20-30 solved problems in each section that span from easy single objective questions to difficult multistep analysis problems Each section of the workbook contains key points that highlight major features of the topic to remind students of what they need to apply to solve problems in the topic area Key Features Includes a visual map that shows how all the equations used in thermodynamics are connected and how they are derived from the three major energy laws Acts as a guide in deriving the correct solution to a problem Illustrates the questions students should ask themselves about the critical features of the concepts to solve problems in physical chemistry Can be used as a stand alone product for review of Thermodynamics questions for major tests **Problems in Applied Thermodynamics** Colin Bodsworth, A. S. Appleton, 1965 **Student Self-study Guide for Thermodynamics** Jack Philip Holman, 1974

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