# CECIL CONVENT SCHOOL

## PERIODIC ASSESSMENT-2

#### Class 11 - Chemistry

Time Allowed: 1 hour

General Instructions:

Maximum Marks: 20

TAIL

111

121

121

131

- 1. There are 11 questions in this question paper.
- SECTION A Q. No. 1 to 5 are very short answer questions carrying 1 mark each.
- 3. SECTION B Q. No. 6 to 8 are short answer questions carrying 2 marks each.
- 4. SECTION C- Q. No. 9 to 11 is case based question carrying 3 marks each.
- 5. All questions are compulsory.
- 6. Attempt questions in order.
- 1. Water can be lifted into the water tank at the top of the house with the help of a pump. Then why is it not considered to be spontaneous?
- Neither q nor W is a state function but q + W is a state function. Explain why?
- 3. What is the enthalpy change for an adiabatic process? [1]
- 4. State the law of chemical equilibrium. . [1]
- 5. A tank is full of water. Water is coming in as well as going out at same rate. What will happen to level of water in a tank? What is name given to such state?
- 6. Calculate the standard enthalpy of formation of CH<sub>3</sub>OH(I) from the following data:
  - i.  $CH_3OH(1) + \frac{3}{2}O_2(g) \longrightarrow CO_2(g) + 2H_2O(1)$ ;  $\Delta_r H^{\odot} = -726 \text{ kJmol}^{-1}$
  - ii.  $C_{(graphite)} + O_2(g) \longrightarrow CO_2(g)$ ;  $\Delta_e H^{\odot} = -393 \text{ k/mol}^{-1}$
  - iii.  $H_2(g) + \frac{1}{2}O_2(g) \longrightarrow H_2O(1); \Delta_f H^\odot = -286 \text{ kJmol}^{-1}$
- The equilibrium constant for a reaction is 10. What will be the value of ΔGΘ? R = 8.314JK<sup>-1</sup>mol<sup>-1</sup>, T = 300 K.
- 8. Write the equilibrium constant (K<sub>c</sub>) expression for the following reactions.
  - (i)  $Cu^{2+}(aq) + 2Ag(s) = Cu(s) + 2Ag^{+}(aq)$
  - (ii)  $4HCl(g) + O_2(g) = 2Cl_2(g) + 2H_2O(g)$
- 9. i. Why standard entropy of an elementary substance is not zero whereas standard enthalpy of formation is taken as zero?
  - ii. Under what conditions will the reaction occur, if
    - a, both  $\Delta H$  and  $\Delta S$  are positive
    - b. both  $\Delta H$  and  $\Delta S$  are negative
- 10. Dihydrogen gas is obtained from the natural gas by partial oxidation with steam as per following endothermic reaction:

$$CH_4(g) + H_2O(g) \rightleftharpoons CO(g) + 3H_2(g)$$

a. Write the expression for Kp for the above reaction.

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understanding of the theory and some reasoning ability to resolve These are followed by self assessment questions which test students understanding of the principles of materials science and are designed to quickly cover the subject area of the chapter This edition of Materials Science for Engineers includes an expanded treatment of many materials particulary polymers foams composites and functional materials Of the latter superconductors and magnetics have received greater coverage to account for the considerable development in these fields in recent years New sections on liquid crystals superalloys and organic semiconductors have also been added to provide a comprehensive overview of the field of materials S. Chand's ICSE Chemistry Book II For Class X (2021 Edition) R.D.MADAN, S Chand's ICSE Chemistry for Class X is strictly in accordance with the latest syllabus prescribed by the Council for the Indian School Certificate Examinations CISCE New Delhi The book aims at simplifying the content matter and give clarity of concepts so that the students feel confident about the subject as well as the competitive exams MCAT General Chemistry Review 2023-2024 Kaplan Test Prep, 2022-07-05 Kaplan's MCAT General Chemistry Review 2023 2024 offers an expert study plan detailed subject review and hundreds of online and in book practice questions all authored by the experts behind the MCAT prep course that has helped more people get into medical school than all other major courses combined Prepping for the MCAT is a true challenge Kaplan can be your partner along the way offering guidance on where to focus your efforts and how to organize your review This book has been updated to match the AAMC s guidelines precisely no more worrying about whether your MCAT review is comprehensive The Most Practice More than 350 questions in the book and access to even more online more practice than any other MCAT general chemistry book on the market The Best Practice Comprehensive general chemistry subject review is written by top rated award winning Kaplan instructors Full color 3 D illustrations from Scientific American charts graphs and diagrams help turn even the most complex science into easy to visualize concepts All material is vetted by editors with advanced science degrees and by a medical doctor Online resources including a full length practice test help you practice in the same computer based format you ll see on Test Day Expert Guidance High yield badges throughout the book identify the topics most frequently tested by the AAMC We know the test The Kaplan MCAT team has spent years studying every MCAT related document available Kaplan's expert psychometricians ensure our practice questions and study materials are true to the test MCAT General Chemistry Review 2026-2027 Kaplan Test Prep, 2025-07-08 Kaplan s MCAT General Chemistry Review 2026 2027 offers an expert study plan detailed subject review and hundreds of online and in book practice questions all authored by the experts behind Kaplan's score raising MCAT prep course Prepping for the MCAT is a true challenge Kaplan can be your partner along the way offering guidance on where to focus your efforts and how to organize your review This book has been updated to match the AAMC s guidelines precisely no more worrying about whether your MCAT review is comprehensive The Most Practice More than 350 questions in the book and access to even more online more practice than any other MCAT general chemistry book on the market The Best Practice Comprehensive general chemistry subject review is

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  - Chapter 2: Essential Elements of Periodic Assessment 2 Chemistry Answers
  - Chapter 3: Periodic Assessment 2 Chemistry Answers in Everyday Life
  - Chapter 4: Periodic Assessment 2 Chemistry Answers in Specific Contexts
  - ∘ Chapter 5: Conclusion
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- 4. In chapter 3, this book will examine the practical applications of Periodic Assessment 2 Chemistry Answers in daily life. This chapter will showcase real-world examples of how Periodic Assessment 2 Chemistry Answers can be effectively utilized in everyday scenarios.
- 5. In chapter 4, this book will scrutinize the relevance of Periodic Assessment 2 Chemistry Answers in specific contexts. This chapter will explore how Periodic Assessment 2 Chemistry Answers is applied in specialized fields, such as education, business, and technology.
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