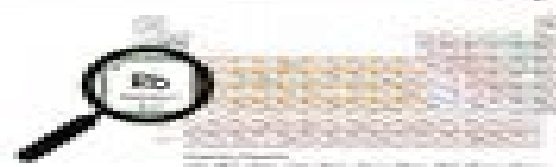


Periodic Table: Scavenger Hunt



Read the information below and answer the questions using the periodic table on page 2.

The **periodic table** is a graphical representation of all known chemical elements. Each box on the periodic table contains information about an element. The labeled image below shows the box on the periodic table for the element hydrogen.



The elements are ordered by their atomic number. Elements in the periodic table are also grouped by their chemical properties. The periodic table is color coded based on these properties.

Use the periodic table on page 2 to find the element or elements that match each description.

1. The **alkali metal** with the atomic number 19. _____
2. The **nonmetal** with an atomic mass of 32.06 _____
3. The **other metal** with an atomic mass of 26.98 _____
4. The **metalloid** with the atomic number 32 _____
5. The **noble gas** with the smallest atomic mass _____
6. The **transition metal** with the chemical symbol Au _____
7. The **lanthanide** with the atomic number 60 _____
8. The **actinide** with an atomic mass of 232.04 _____
9. The **transition metal** with the smallest atomic mass and an atomic number between 27 and 30 _____
10. The two **alkaline earth metals** with chemical symbols that start with the letter "B" a. _____ b. _____

Periodic Table Trends Scavenger Hunt

Eric R. Scerri



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Differentiation That Really Works Cheryll M. Adams, Rebecca L. Pierce, 2021-09-23 Differentiation That Really Works Science provides time saving tips and strategies from real teachers who teach science in grades 6 12 These teachers not only developed the materials and used them in their own classes but they also provided useful feedback and comments about the activities The strategies included in the book are tiered lessons cubing graphic organizers exit cards learning contracts and choice boards Every strategy includes directions and offers opportunities for differentiation Grades 6 12 [Current Index to Journals in Education](#) ,1985 *The Software Encyclopedia* 2001 ,2001 **Periodic Table, The: Past, Present, And Future** Geoff Rayner-canham, 2020-08-04 This is an absolutely wonderful book that is full of gems about the elements and the periodic table All in all the book is highly recommended to philosophers of chemistry As philosophers we have a natural tendency to concentrate on generalities and not to get too involved in the specifics and the details Above all else this new book reminds us that such an approach needs to be tempered by a detailed knowledge of the exceptions and features that go against the simplified generalities which we so cherish Read Full Review Eric Scerri Foundations of Chemistry Many questions are dealt with in a clearly written way in this stimulating and innovative book The reader will quickly become interested in the subject and will be taken on tour through this Periodic Table in a very readable way both for students and teachers The number of illustrations is good and clear This book is indeed unique and quite thought provoking This book is highly recommended for students teachers researchers and not only chemists Geologists biochemist and also physicists will find it very interesting to read Read Full Review Chemistry International That fossilized chart on every classroom wall isn't that The Periodic Table Isn't that what Mendel ever devised about a century ago No and No There are many ways of organizing the chemical elements some of which are thought provoking and which reveal philosophical challenges Where does hydrogen belong Can an element occupy more than one location on the chart Which are the Group 3 elements Is aluminum in the wrong place Why is silver like thallium Why is vanadium like molybdenum Why does gold form an auride ion like a halide ion Does an atom know if it is a non metal or metal Which elements are the metalloids Which are the triels So many questions In this stimulating and innovative book the Reader will be taken on a voyage from the past to the present to the future of the Periodic Table This book is unique This book is readable This book is thought provoking It is a multi dimensional examination of patterns and trends among the chemical elements Every reader will discover something about the chemical elements which will provoke thought and a new appreciation as to how the elements relate together [Periodic Table Advanced](#) BarCharts, Inc, Mark Jackson, 2014-12-31 The ultimate reference tool and lab partner for any student of science durably laminated and designed to fit as much info as possible in this handy 6 page format Separate property tables are broken out for the ease of locating trends while studying and working while other pages offer essential notes about the table's organization and history Consistently a best seller since its first creation [Chemistry and Our Universe](#) , Return to the

periodic table introduced in Lecture 1 to practice predicting properties of elements based on their electronic structure Then witness what happens when three different alkali metals react with water Theory forecasts a pronounced difference in the result Is there

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artworks reveal aspects of their atomic structure that explain why they behave the way they do For example why is gold so malleable and diamond so incredibly hard How can bismuth levitate and why does cobalt make paint such a vivid blue If you want to unlock the secrets of matter itself open *The Periodic Table* *The Periodic Table* Eric R. Scerri,2019-07-25 The periodic table of elements first encountered by many of us at school provides an arrangement of the chemical elements ordered by their atomic number electron configuration and recurring chemical properties and divided into periodic trends In this Very Short Introduction Eric R Scerri looks at the trends in properties of elements that led to the construction of the table and shows how the deeper meaning of the table s structure gradually became apparent with the development of atomic theory and in particular quantum mechanics which underlies the behaviour of all of the elements and their compounds This new edition publishing in the International Year of the Periodic Table celebrates the completion of the seventh period of the table with the ratification and naming of elements 113 115 117 and 118 as nihonium moscovium tennessine and oganesson Eric R Scerri also incorporates new material on recent advances in our understanding of the origin of the elements as well as developments concerning group three of the periodic table ABOUT THE SERIES The Very Short Introductions series from Oxford University Press contains hundreds of titles in almost every subject area These pocket sized books are the perfect way to get ahead in a new subject quickly Our expert authors combine facts analysis perspective new ideas and enthusiasm to make interesting and challenging topics highly readable *The Periodic Table* Mary Griffin,2018-12-15 The periodic table is the way scientists have organized the known elements but up and coming scientists can learn much from the periodic table as well Your readers will learn how to read the periodic table including what an element s atomic number means and what the rows and columns signify They ll gather facts about common and interesting elements and even discover how they can predict how elements might chemically react from their position in the table Fact boxes and images add additional information to the accessible text *Trends in the Periodic Table* Open University. S25- Course Team,1972 **Periodic Table Advanced** BarCharts, Inc,2014 *Mastering the Periodic Table* Linda Trombley,2000 **The Periodic Table** Geoffrey Rayner-canham,2020 *The Periodic Table I* D. Michael P. Mingos,2020-02-05 As 2019 has been declared the International Year of the Periodic Table it is appropriate that Structure and Bonding marks this anniversary with two special volumes In 1869 Dmitri Ivanovitch Mendeleev first proposed his periodic table of the elements He is given the major credit for proposing the conceptual framework used by chemists to systematically inter relate the chemical properties of the elements However the concept of periodicity evolved in distinct stages and was the culmination of work by other chemists over several decades For example Newland s Law of Octaves marked an important step in the evolution of the periodic system since it represented the first clear statement that the properties of the elements repeated after intervals of 8 Mendeleev s predictions demonstrated in an impressive manner how the periodic table could be used to predict the occurrence and properties of new elements Not all of his many predictions proved to be valid but the discovery of scandium

gallium and germanium represented sufficient vindication of its utility and they cemented its enduring influence Mendeleev's periodic table was based on the atomic weights of the elements and it was another 50 years before Moseley established that it was the atomic number of the elements that was the fundamental parameter and this led to the prediction of further elements Some have suggested that the periodic table is one of the most fruitful ideas in modern science and that it is comparable to Darwin's theory of evolution by natural selection proposed at approximately the same time There is no doubt that the periodic table occupies a central position in chemistry In its modern form it is reproduced in most undergraduate inorganic textbooks and is present in almost every chemistry lecture room and classroom This first volume provides chemists with an account of the historical development of the Periodic Table and an overview of how the Periodic Table has evolved over the last 150 years It also illustrates how it has guided the research programmes of some distinguished chemists

Science Comics: The Periodic Table of Elements Jon Chad, 2023-03-14 Step out of your element with Science Comics The Periodic Table of Elements the latest volume of First Second's nonfiction graphic novel series A book of fun chemistry experiments has fallen into the wrong hands Only Mel can use her knowledge of the periodic table to put an end to a maniacal madman's evil schemes The periodic table helps us quickly understand the 118 elements those tiny substances that make up everything in the world By using the periodic table we can recognize how these building blocks behave find trends and patterns in the universe and make predictions about elements that haven't been discovered yet Join us in learning about the periodic table and maybe the next big discovery will be yours *Periodic Table* Adrian Dingle, 2022 Beginning with a concise history of chemistry scientific pioneers and the creation of the first periodic table this comprehensive guide then launches into a visual tour of each individual element Along the way you'll find out where each element comes from and what it is used for explained clearly and simply for young readers Explore elements such as nitrogen and oxygen and learn why they are essential to our survival See how precious gold protects astronauts in space and what makes the metal mercury so unusual Find out about synthetic elements created in labs which the smartest chemists are still busy figuring out how to use

Understanding the Periodic Table Jane P. Gardner, 2018 Introduces the periodic table of the elements

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- Chapter 4: Periodic Table Trends Scavenger Hunt in Specific Contexts
- Chapter 5: Conclusion

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This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Periodic Table Trends Scavenger Hunt.

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