

Periodic Table Puzzles

Name _____

Fictitious symbols are used for the first 18 elements in the periodic table. Use the clues below to write the fictitious symbol in the appropriate spot on the periodic table provided. Symbols for real elements do not represent those elements. HINT: You do not have to complete each clue in order.

1	2	13	14	15	16	17	18
X		A	T	H P	G	Q	L
U	R	AI	DP	Z	E	K	W
J	M						V

Clue #1 U and J are alkali metals. J has more energy levels than U.

Clue #2 I has 4 valence electrons on the ^{2nd} 3rd energy level.

Clue #3 M is a metal in period 3 with 2 valence electrons

Clue #4 X has one proton in its nucleus

Clue #5 Q has 2 energy levels and is a halogen

Clue #6 L is the only noble gas that doesn't have 8 valence electrons

Clue #7 A is the only metalloid in the 13th group.

Clue #8 I has 3 valence electrons on the third energy level

Clue #9 V and W are un-reactive elements. V has more electrons than W.

Clue #10 E and G have a common ion of -2 (E^{-2} , G^{-2}). E has 16 neutrons.

Clue #11 R has an nuclear symbol of 3_4R

Clue #12 K has one less electron than V

Clue #13 D and Z are in the same period as J. However D is a metalloid and Z is a nonmetal.

Clue #14 H has 7 neutrons

Periodically Puzzling Answers

Mark P. Kritzman



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Table of Contents Periodically Puzzling Awnsers

1. Understanding the eBook Periodically Puzzling Awnsers
 - The Rise of Digital Reading Periodically Puzzling Awnsers
 - Advantages of eBooks Over Traditional Books
2. Identifying Periodically Puzzling Awnsers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Periodically Puzzling Awnsers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Periodically Puzzling Awnsers
 - Personalized Recommendations
 - Periodically Puzzling Awnsers User Reviews and Ratings
 - Periodically Puzzling Awnsers and Bestseller Lists
5. Accessing Periodically Puzzling Awnsers Free and Paid eBooks
 - Periodically Puzzling Awnsers Public Domain eBooks
 - Periodically Puzzling Awnsers eBook Subscription Services
 - Periodically Puzzling Awnsers Budget-Friendly Options
6. Navigating Periodically Puzzling Awnsers eBook Formats

- ePub, PDF, MOBI, and More
- Periodically Puzzling Awnsers Compatibility with Devices
- Periodically Puzzling Awnsers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Periodically Puzzling Awnsers
 - Highlighting and Note-Taking Periodically Puzzling Awnsers
 - Interactive Elements Periodically Puzzling Awnsers
- 8. Staying Engaged with Periodically Puzzling Awnsers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Periodically Puzzling Awnsers
- 9. Balancing eBooks and Physical Books Periodically Puzzling Awnsers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Periodically Puzzling Awnsers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Periodically Puzzling Awnsers
 - Setting Reading Goals Periodically Puzzling Awnsers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Periodically Puzzling Awnsers
 - Fact-Checking eBook Content of Periodically Puzzling Awnsers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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