

CHEMISTRY

2018

Time: 2 Hours 40 Minutes

Marks: 68

SECTION 'B' (SHORT-ANSWER QUESTIONS)

NOTE: Answer 10 questions from this section(40)

INORGANIC CHEMISTRY

- 2.(i) Define Period and Group. Give the valence shell configurations of Alkali metals, Carbon family, Halogens and Noble gases.
- (ii) What are Isotopes? Explain the different isotopes of hydrogen.
- (iii) Refer to the list of the given compounds:

Compound	A	B	C	D
Specific name	Epsom Salt	Plaster of paris	Bleaching powder	Baking soda

- Write the formulae of A and D
 - The equation for the preparation of B.
 - Give only one common use of B.
 - Give the equation for the reaction of C with HCL.
- (iv) Give scientific reasons for any four of the following:
- Diamond is non-conductor of electricity.
 - H_2O and NH_3 act as ligands but H_3O^+ & NH_4^+ do not.
 - Blue crystals of Copper sulphate turn white on heating.
 - Alkali metals are powerful reducing agents.
 - Alkaline earth metals are more strongly hydrated than alkali metals.
- (v) Give the refining of Aluminium by Hoope's electrolytic method (no diagram needed).
- (vi) Write I.U.P.A.C. names of the following complexes:
- $[\text{Pt}(\text{NH}_3)_2 \text{Cl}_4]$
 - $\text{NH}_4[\text{Cr}(\text{SCN})_4(\text{NH}_3)_2]$
 - $[\text{Cu}(\text{CN})_4]^{-2}$
 - $[\text{Co}(\text{H}_2\text{N} - \text{CH}_2 - \text{CH}_2 - \text{NH}_2)_3] (\text{NO}_3)_3$
- (vii) Discuss any two of the following properties of Transition metals.
- Magnetic properties
 - Colour
 - Formation of complexes
- OR
- What is Corrosion? Write its causes. Give different methods for the prevention of metals from Corrosion.
- (viii) Give complete and balanced equations for any four of the following:
- Action of NaOH on Carbon monoxide.
 - Action of Concentrated HNO_3 on Sulphur.

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David Mahony



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