

Chapter Quiz

15-1 Which statement is correct about water's solid state? **100%**

- ☐ A. The expansion that occurs during freezing is due to the fact that ice is lighter than liquid water.
- ☒ B. Hydrogen bonds are responsible for the expansion.
- ☐ C. Water expands.

100%

☐ D. Hydrogen bonds are

100%

- ☐ A. responsible for the expansion that occurs during freezing.
- ☐ B. the reason that ice floats on liquid water.

100%

☐ C. responsible for the expansion that occurs during freezing.

100%

☐ D. not related to the expansion that occurs during freezing.

100%

- ☐ A. Hydrogen bonds following the expansion of water are responsible for the expansion that occurs during freezing.
- ☐ B. Water expands.

☐ C. Water

☐ D. contracts.

☐ E. None

☐ A. Hydrogen bonds following the expansion of water are responsible for the expansion that occurs during freezing.

☐ B. Water expands.

☐ C. Water

☒ D. contracts.

☐ E. None

100% correct! Congratulations! You have passed this chapter's quiz.

100%

☐ A. The expansion that occurs during freezing is due to the fact that ice is lighter than liquid water.

☐ B. Hydrogen bonds are responsible for the expansion that occurs during freezing.

☐ C. Water expands.

☐ D. Hydrogen bonds following the expansion of water are responsible for the expansion that occurs during freezing.

☐ E. None

☐ A. Hydrogen bonds following the expansion of water are responsible for the expansion that occurs during freezing.

Water And Aqueous Systems Chapter Test A Answers

**Dana W. Mayo, Ronald M. Pike, David C.
Forbes**

Water And Aqueous Systems Chapter Test A Answers:

Water in Crystalline Hydrates Aqueous Solutions of Simple Nonelectrolytes Felix Franks, 2013-04-18 vi the information collected and discussed in this volume may help toward the achievement of such an objective I should like to express my debt of gratitude to the authors who have contributed to this volume Editing a work of this nature can strain long established personal relationships and I thank my various colleagues for bearing with me and responding sooner or later to one or several letters or telephone calls My special thanks once again go to Mrs Joyce Johnson who bore the main brunt of this seemingly endless correspondence and without whose help the editorial and referencing work would have taken several years F FRANKS Biophysics Division Unilever Research Laboratory Colworth Welwyn Colworth House Sharnbrook Bedford January 1973 Contents Contents of Volume 1 xv Contents of Volume 3 xvi Contents of Volume 4 xvii Chapter 1 The Solvent Properties of Water F Franks 1 Water the Universal Solvent the Study of Aqueous Solutions 2 Aqueous Solutions of Nonelectrolytes 5 2 1 Apolar Solutes 6 2 2 Polar Solutes 19 2 3 Ionic Solutes Containing Alkyl Residues Apolar Electrolytes 38 3 Aqueous Solutions of Electrolytes 42 3 1 Single Ion Properties 42 3 2 Ion Water Interactions 43 3 3 Interionic Effects 47 4 Complex Aqueous Mixtures 48 Chapter 2 Water in Stoichiometric Hydrates M Falk and O Knop 1 Introduction 55 2 Symmetry and Types of Environment of the H₂O Molecule 2 in Crystals 57 vii Contents viii 2 1 Site Symmetry 57 Excel Preliminary Chemistry C. M. Roebuck, 2003 Introduction to Chemistry Martha J. Gilleland, Rebecca Williams, 1986

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