

Experiment 35

Advance Study Assignment: Spot Tests for Some Common Anions

1. Each of the observations in the following list was made on a different solution. Given the observations, state which ion studied in this experiment is present. If the test is not definitive, indicate that with a question mark.
- A. Addition of 6 M NaOH and Al to the solution produces a vapor that turns red litmus blue.
Ion present:
 - B. Addition of 6 M HCl produces a vapor with an acrid odor.
Ion present:
 - C. Addition of 6 M HCl produces an effervescence.
Ion present:
 - D. Addition of 6 M HNO₃ plus 1.0 M BaCl₂ produces a precipitate.
Ion present:
 - E. Addition of 6 M HNO₃ plus 0.10 M AgNO₃ produces a precipitate.
Ion present:
 - F. Addition of 6 M HNO₃ plus 0.5 M (NH₄)₂MoO₄ produces a precipitate.
Ion present:
2. An unknown containing one or more of the ions studied in this experiment has the following properties.
- A. No effect is observed on addition of 6 M HNO₃.
 - B. A white precipitate is observed on addition of 1.0 M BaCl₂ to the solution from Part A.
 - C. No effect is observed on addition of 0.10 M AgNO₃ to the solution from Part A.
 - D. A yellow precipitate is observed on addition of (NH₄)₂MoO₄ to the solution from Part A.

On the basis of this information, which ions are present, which are absent, and which are in doubt?

Present

Absent

In doubt

Spot Tests Advance Study Assignment

Rick Donnelly



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OAR Quarterly Index of Current Research Results United States. Air Force. Office of Aerospace Research, 1967

OAR Cumulative Index of Research Results, 1967

OAR Index of Research Results, 1967

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