

1. Which number has *the most* significant figures?

300.00 5 sig figs

all zeros in this number are significant as they come at the end of the number with a decimal point

5.78 3 sig figs

0.000019 2 sig figs

32 000 000 2 sig figs

2. Which of the following *correctly presents* the result of the calculations:

$$94.300 \text{ L} + 281.7 \text{ L} - 0.27 \text{ L}$$

375.7 L

the answer should have one digit after decimal, the rule for addition/subtraction

3. The measurement 0.00000284 mg, expressed correctly using exponential notation:

$2.84 \times 10^{-6} \text{ mg}$

The power is negative as the original number is small

4. Which of the following numbers was *correctly rounded* to **three** significant figures?

361.892 to 362 this number is correctly rounded and has 3 sig figs

80.0862 to 81.0 - rounding is incorrect

0.05114 to 511 - rounding is incorrect

9.02438 to 9.024 this number is correctly rounded but has 4 sig figs

5. A pack of Paprika weighs 3.76 oz. Convert this mass into grams? 1 oz = 28.40 g

107 g

$3.76 \text{ oz} \times \frac{28.40 \text{ g}}{1 \text{ oz}} = 106.784 \text{ g}$; the number should be rounded to have 3 sig figs as in a number of oz given

Spring Semester Senior Chemistry Exam Answers

Diane M. Bunce, Cinzia M. Muzzi



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Measurable Outcomes of Individual Laboratory Work in High School Chemistry Ralph Ellison Horton, 1928

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