



2. In solvent like H_2O



Halohydrin formation

- + It is a **regioselective** reaction
- + The nucleophilic attack of the water occurs at the more substituted carbon atom.

2. With Alkynes

- + It is an **anti-addition** reaction:
- 1. In solvent like CCl_4, CH_2Cl_2 etc



Tetrabromide formation

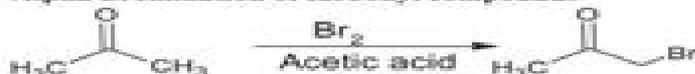
3. With Aromatic Compounds



4. Photo halogenation of alkanes

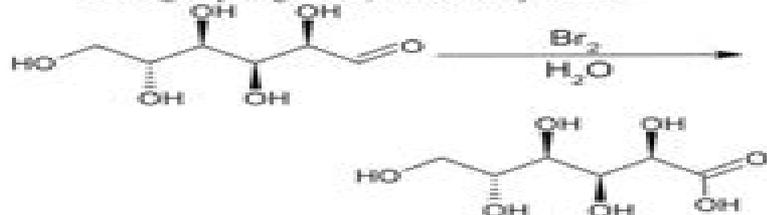


5. Alpha Bromination of carbonyl compounds



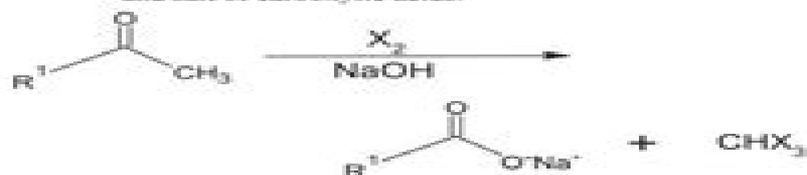
6. Oxidation of Monosaccharides

1. Bromine water is a mild oxidizing agent
2. selectively oxidizes the aldehyde group (for example CHO group of glucose) to a carboxylic acid.



7. Haloform Reaction

1. converts methyl ketones to Haloform (CHX_3 ($X=Cl, Br, I$)) and salt of carboxylic acids.

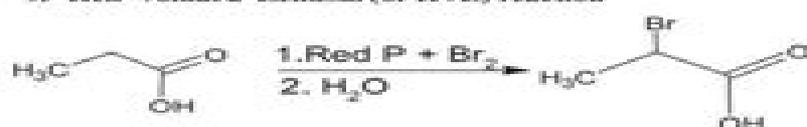


8. The Hoffmann Bromamide degradation

1. It converts primary amide to a primary amine with one less carbon atom.



9. Hell-Volhard-Zelinski (or HVZ) reaction



5. CN^-

Cyanide Ion and Nitrile Group

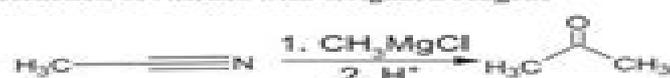
1. Conversion of Nitriles to a Primary Amine



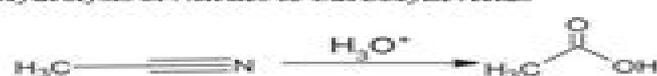
2. Conversion of Nitriles to Aldehydes



3. Reaction of Nitriles with Grignard reagent



4. Hydrolysis of Nitriles to Carboxylic Acids



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